

Analysis of groundwater from deep boreholes in Fjällveden

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ANALYSIS OF GROUNDWATER FROM DEEP BOREHOLES IN FJÄLLVEDEN

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IVL, Swedish Environmental Research Institute Stockholm, Sweden 1983-03-29

This report concerns a study which was conducted for SKBF/KBS. The conclusions and viewpoints presented in the report are those of the author(s) and do not necessarily coincide with those of the client.

A list of other reports published in this series during 1983 is attached at the end of this report. Information on KBS technical reports from 1977-1978 (TR 121), 1979 (TR 79-28), 1980 (TR 80-26), 1981 (TR 81-17) and 1982 (TR 82-28) is available through SKBF/KBS.

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SUMMARY

Groundwater from three boreholes at an investigation site in Fjällveden has been sampled and analyzed. This is part of a larger programme of geological, geophysical and hydrogeological investigations aimed at finding a suitable site for a high level radioactive waste repository.

Four water-bearing levels in each of the boreholes FJ2 and FJ4, and two levels in borehole FJ8, have been sampled. The sampling depths range from 106 down to 562 m. Prior to sampling, the water-bearing section is isolated between packer sleeves. The water is then pumped to the surface where sensitive parameters such as redox potential, pH, sulphide and oxygen content are measured electrochemically on the flowing water in a system isolated from the air. Water, filter and gas samples are sent to several laboratories for further analysis.

The present report is a presentation of the results of the groundwater analyses. The reliability of the results is discussed but there is no evaluation in relation to geology and hydrogeology. This report presents the basic results from the groundwater analyses to be further evaluated by experts in different fields.

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INTRODUCTION

Groundwater from three boreholes in Fjällveden has been sampled and analyzed. The sampling levels were chosen on the basis of hydrological measurements in the boreholes. Sampling was carried out by Sveriges Geologiska AB. Personnel from IPK and VIAK performed the field measurements and sampling. Several laboratories participated in the water analysis (see below). Supplementary sampling in additional boreholes in Fjällveden is in progress. The results will be reported at a later time.

The boreholes were pumped out with a gas lift prior to the hydrological investigation and immediately before the equipment for water sampling was lowered to the first sampling level. On the latter occasion, three such gas lift pumpings - called mammoth pumpings were performed in a row. Iodide (NaI, 0.01 mmol/1) was added to the drilling water during drilling to enable any residual drilling water to be traced in the groundwater. The selected fracture zones were sealed off by packers spaced at a distance of 2.7 m.

Sampling took place from the middle of June to the middle of October, 1982. Four levels were sampled in each of the boreholes FJ2 and FJ4, and two levels in FJ8. Depth is given both as vertical depth (depth) and/or borehole length (length). It is always vertical depth that is given, unless otherwise specified in the table heading.

The work was carried out in periods of 14 days - lowering and start of the sampling pump, 4 days off and 9 days field measurements. When the supply of water is adequate, approximately 4.5 cubic metres of water can be pumped up during one period.

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The present report consists primarily of a presentation of the results of the groundwater analyses. The reliability of the results is subjected to some scrutiny. There is, however, no evaluation in relation to geology or hydrology. The material in this report will be further analyzed by experts in different fields.

The geology and hydrology of the test area will be described in KBS TR 83-52.

A general description of the chemistry of groundwater from great depths in granite and gneiss has previously been published by G Jacks (KBS TR 88). The same author has also described the chemistry of groundwater in Blekinge (KBS TR 79-07), where Sternö is situated. The results of analyses of the groundwater from the Finnsjö area are presented in KBS TR 82-23 and from Gideå in KBS TR 83-17. A geochemical evaluation of fracture minerals and the relation between fracture minerals and groundwater composition in Finnsjön has been published by E-L Tullborg and co-workers (KBS TR 82-20). The relationship between pH and carbonate content in deep groundwaters has been discussed by Bert Allard in KBS TR 82-25. The expected groundwater composition and its importance for the final storage of radioactive waste have been discussed in KBS TR 90 and the final report KBS-2 Volume 2 (Handling and Final Storage of Unreprocessed Spent Nuclear Fuel, Technical Volume). Hydrology and groundwater age are also taken up in the latter report.

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SAMPLING

2.1 Drilling water

Some of the drilling water that is used in the core drilling of the sampling holes can penetrate into rock fractures and contaminate the groundwater there. In order to get rid of the drilling water to as great an extent as possible prior to sampling, three mammoth pumpings (using nitrogen gas) are performed immediately prior to installation of the sampling equipment. The sampling equipment (2.2) is lowered to the chosen level and allowed to work for 3-4 days before sampling begins. Periodic sampling during a period of 8-9 days makes it possible to follow changes in the composition of the water that might derive from drilling water or some other source of contamination. The changes may also be due to the incursion of water from different aquifers.

Water from percussion-drilled boreholes has been used as drilling water. The drilling water was filtered through mechanical filters of cellulose (18 CMC 3-2), which are supposed to retain particles with diameters larger than $5 \,\mu$ m. The filtered drilling water was then analyzed. The results for the boreholes in question are presented at the bottom of the respective tables (tables 5-8).

Before the drilling water was used, it was "marked" by the addition of sodium iodide (0.01 mmol/l in the drilling water) in order to make sure that it would be possible to trace any drilling water that may have contaminated the sampled water. Unfortunately, it has been found that both the drilling water and sampled groundwater can have considerable contents of natural iodide (KBS TR 83-17). Chloride, for example, is always accompanied by small quantities of bromide and iodide. If we assume 3×10^{-4} mg I per mg Cl, which is a value that has been observed in connection with water sampling at Stripa, chloride concentrations of 100-200 mg/l correspond to iodide concentrations of 0.03-0.06 mg/l, which in turn would be erroneously interpreted as a drilling water contamination of 2-5%. The drilling water contamination given for FJ2, 506 m should for this reason be 2-4% too high, and the drilling water contamination in the first sample (No. 231) in FJ4, 420 m can probably be explained by the high salt content. The drilling waters used in Fjällveden were not analyzed for iodide prior to marking, which brings an additional uncertainty to the drilling water contaminations given in table 8.

2.2 Sampling equipment

The boreholes are core-drilled with a diameter of 56 mm. The 2.7 m long sampling zone is sealed off by rubber packers that are expanded

to a pressure that is 0.8-1 MPa above the groundwater pressure. The sampling pump is positioned immediately above the sealed-off zone. The intake is from the upper part of the zone. The pump, which is made of steel and furnished with teflon seals, has a maximum capacity of 0.12 l/stroke, equivalent to 330 l/d. The water flow available for sampling decreases if the flow of water in the measured rock is lower than the capacity of the pump. A schematic illustration of the sampling pump and packers is shown in Fig. 1.

The water is pumped up through teflon-lined iron pipes to a test chamber of stainless steel on the ground surface. The test chamber is equipped with electrodes and measuring cells for recording pH, Eh (glassy carbon), Eh (platinum), pS and the oxygen content and conductivity of the water. The test chamber is also equipped with valves for the extraction of water samples.

As of September, the equipment has been augmented with an extra measuring chamber of glass with two Eh electrodes of the same kind as those in the test chamber (glassy carbon and platinum). In order to give these electrodes plenty of time to stabilize, they are not calibrated and they are only cleaned when the level is changed.

2.3 Procedure

Work at the boreholes is conducted in periods of 14 days. On the last day of a period, the sampling equipment is set up at the level that is to be sampled during the following period, and the sampling pump is started. The personnel then go home for 4 days off while the sampling pump works. Sampling then proceeds for 10 consecutive days. There have been some disruptions in the programme, as is evident from the notes to the table in 2.4.

2.4 Water flow during sampling

The water flow at different levels is shown in the table below. The pumped-up water volume has also been calculated. "Prior" refers to

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the approximate quantity of water pumped up from the level before sampling started, and "Sampl." to the quantity of water during the sampling period.

BH	Depth m	Mean flow 1/day	Prior m ³	Samp1. m3	Total m ³	Notes
FJ2	106	270	0.2	1 5	1 7	-1
.02	293	245	0.2	1.5 2.2	1.7	1
	409	265				2
				1.6		3
	506	190	1.1	1.7	2.8	
FJ4	131	315	1.9	2.5	4.4	
	272	320	1.9	2.6	4.5	
	349	335	2.0	2.7	4.7	
	420	260	0.2	1.6	1.8	4
FJ8	402	250	0.0	1.8	1.8	5
	562	145	0.9	1.3	2.2	2
	£					

Note 1 Due to delays at the borehole, sampling started the day after lowering.

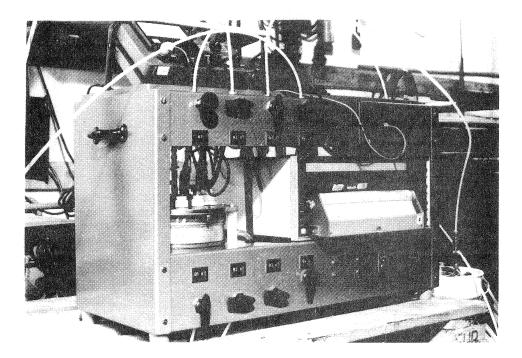
- Note 2 The pump ran for about seven days prior to sampling, but unfortunately there was no pressure in the packers when the personnel returned after their leave. It is therefore probable that water leaked into the sampling zone.
- Note 3 Due to difficulties in connection with lowering, the pump was not started until 14 h before the start of the measuring period. The pressure in the packers was lost again in that time, so water may have leaked into the sampling zone.
- Note 4 Here again, difficulties were encountered in connection with lowering. The sampling zone, which is only sealed off by a single packer, goes from the bottom of the borehole at 585 m to 420 m. The sampling pump started the day before the measurements.

Note 5 The pump was started 2.5 hours before the readings began. The 1st sampling was done 24 hours later.

3 FIELD MEASUREMENTS

The field measurements are recorded in tables 1-3, the E^{O} values obtained from the field calibrations in table 4 and the field values obtained during water sampling in table 5.

The photograph below shows the field equipment with test chamber, measuring equipment and valves for water sampling.



3.1 Calibrations

Each calibration has been assigned a calibration number. Measurements made after the calibration have the same number.

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All calibration solutions are freshly prepared from concentrate and deaerated distilled water, except the one used for calibration of the conductivity cell.

Three buffer solutions with pH 4, 7 and 10 are used for calibration of the pH electrode. Quinhydrone is added to the buffer solutions with pH 4 and 7 for simultaneous calibration of the Eh electrodes. Concentrate diluted to solutions that are 0.01 and 0.05 molar with respect to sulphide ion is used for calibration of the sulphide electrode. The E^{0} values obtained are presented in table 4.

The oxygen probe is calibrated against air-saturated distilled water and the conductivity cell against 0.01 and 0.1 mol/l KCl.

3.2 Measurement results

Each measuring occasion is identified with a calibration number, borehole designation, length of borehole (core length) as well as day and time. The vertical depths corresponding to the borehole lengths are shown in the table below.

Bore-	Length	Depth	Bore-	Length	Depth	Bore-	Length	Depth
hole	m	m	hole	m	m	hole	m	m
FJ2	123 342 483 605	106 293 409 506	FJ4	151 317 410 496	131 272 349 420	FJ8	470 666	402 562

Owing to the fact that it takes some time after each calibration before the water in the test chamber is representative of the borehole water and before the Eh and pS values have stabilized, no measurement results have been included until at least 24 hours after the calibration - in general, from and including the second day after calibration. Nor are values included that have been obtained from obviously unsuccessful measurements owing to problems encountered in the field. Due to a fault in the measuring equipment at FJ4, pH values were only obtained from the measurements during the first days on the upper level.

Stable measurement results were not obtained from the extra Eh electrodes that were connected before the test chamber until voltmeters with higher internal resistance were procured. The values measured with the extra electrodes are marked "ex" in table 3.

3.3 Temperature measurement

The temperature is measured in the test chamber to permit calibration and calculation of the electrode potentials.

Owing to the relatively slow rate of water turnover in the test chamber, the temperature measured is not representative of the groundwater. It is therefore not reported.

3.4 Measurement of pH, Eh and pS

The system often seems to need to remain undisturbed for about a week before the Eh and pS values are more or less stable. The pH value stabilizes much faster.

The calibrations show that adjustments must be made fairly frequently, probably because a deposit builds up on electrodes and probes. The optimum time between two calibrations would therefore appear to be between 5 and 10 days. The electrodes were usually cleaned when they were lowered to a new level as well as before calibrations 203, 204, 207, 208, 304, 305, 306, 403 and 406.

Due to an amplifier malfunction, almost all pH values for FJ4 are lacking.

3.5 Oxygen measurement with probe

When the oxygen probe was recalibrated, it was found that the measuring error was 5-25% - in the negative direction, as a rule. In seven cases, errors greater than 40% were found, and the probe had to be cleaned and the membrane changed. Values before such a major adjustment are not reported.

It appears as if the oxygen probe has to be recalibrated more frequently than is possible according to 3.4 in order for reliable results to be obtained.

Despite the difficulties encountered in oxygen measurement, the field values often show good correlation with pS; high oxygen content - high pS (low sulphide content).

3.6 Measurement of conductivity

Calibration was only done at the beginning of each level. The conductivity measurements were consistently stable.

4 MAIN COMPONENTS OF THE WATER

Each sampling occasion has been assigned a unique sampling number, which is the same for all samples taken on the same occasion ("Nr." in the tables).

The main components were analyzed by SGAB's water laboratory in Uppsala and by Hydroconsult in Stockholm. Some control analyses were performed by IVL (the Swedish Institute for Water and Air Polution Research) in Stockholm and IMLAB in Linköping. The analysis results are reported in tables 5-8. The table below shows which laboratory carried out the analyses and which method was used. Where there are two analyses of the same parameter, the mean value is reported.

	SGAB	H-con.		SGAB	H-con.
$\mathbf{p}\mathbf{H}$	*	*	HCO ₃	g	Ъ
cond.	*	*	C1 ³	i	h
Turb.		*	F	i	k
Ca	а	Ъ	SO,	i	
Mg	а		PO,4		f
Na	а	С	NO_{3}^{4}		f
K	а	С	NO ³		f
Mn	а	d	NH ²		f
Fe(II)	е		sið,	m	m
Fe-tot	а	·	s ^z	n	
A1	а		I	0	
Cu	а		TOC	р	q
Sr	а			-	-

* Conventional according to Swedish Standard (SS)

a Optical emission with ICP (own design)

b Titrimetric with EDTA according to SS

c Emission with flame

d Atomic absorption

e Colorimetric with o-phenanthroline according to Standard Methods

f Colorimetric according to SS

g Specially designed titrator

h Titration according to Mohr (SS), high concentrations only

i Ion chromatograph, for chloride, low concentrations only

k Potentiometrically according to SS

- m Colorimetrically with methylene blue according to Standard Methods
- n Potentiometrically
- Colorimetrically, higher concentrations also potentiometrically
- p Carbon analyzer from Carlo-Erba
- q Carbon analyzer from Oceanographic

Drilling water was analyzed by SGAB. The analyses are presented at the bottom of tables 5, 6 and 8.

The field values obtained simultaneously with the water sampling and the laboratory analyses of the same parameters are presented in table 5. Note that the field values, which are measured during water sampling, are often not representative of the borehole. Sampling may, for example, have been done immediately after a calibration.

The date is given for both the sampling day and the day the samples were received at the laboratories. In both cases, the date is given with the number of the week followed by the number of the day of the week. (Day 1 is Monday.)

All determinations of metal ions are presented in table 6, while table 8 shows TOC, turbidity and the negative ions that do not contain nitrogen.

The parameters that contain nitrogen are compiled in table 7. The analysis results are given as the concentration of the ion and the concentration of nitrogen.

"Drilling water residue" (table 8) is the percentage fraction of the original iodide concentration in the drilling water (2.1) that has turned up in the groundwater. No correction is made for the water's natural iodide content (2.1).

4.1 Sampling

The water samples were taken through a valve in immediate connection with the test chamber. On each sampling occasion, a new tube, kept well protected from dust, was attached to the valve. The free and untouched end of the tube was inserted down to the bottom of the sample bottle. At least two bottle volumes of sample water were allowed to run over before the bottle was sealed, without any air having been trapped inside. The samples were kept in a refrigerator awaiting transport (4.2). 4.1.1 Unpreserved samples for analysis of negative ions etc. were taken in bottles of borosilicate glass (1 litre) with a ground, filled stopper held in place with a steel clip.

In order to keep the ground surface moist so that no air can leak through, the bottles were stored and transported upside-down.

4.1.2 Preserved samples for analysis of metal ions were taken in acidwashed polyethylene bottles (250 ml).

The samples were preserved with 2.5 ml of concentrated hydrochloric acid (super-pure). The acid was added through a dispenser below the sample surface immediately before the sample bottle was sealed. Hydrochloric acid was chosen as a preservative because iron(II) was to be determined.

4.2 Transport of samples

The samples were packed in specially-made insulated boxes with frozen blocks. The boxes were mailed "express" on the afternoon of the sampling day and delivered to the laboratory the following morning. No water sampling took place on Fridays and Saturdays, since the samples could then not be received by the laboratory until Monday or Tuesday. The Sunday samples were sent together with the Monday samples.

4.3 Sampling levels

Field analyses representative of the borehole are presented in tables 1-3, while the field analyses in table 5 are only representative of the sampling occasion.

Except for the redox-sensitive parameters, the analysis results are, as a rule, strikingly consistent within the levels in FJ2 and FJ8. In FJ2, 106m, as well as in all of FJ4, pronounced trends are found in the concentrations of anions and cations, least pronounced in FJ4, 272m where only iron, phosphate and nitrogen-containing ions change. There is a possibility that water has leaked into the sampling zone due to a pressure loss in the packers immediately before the start of sampling at levels FJ2, 293m and 409m. No appreciable effect of the leakage is noticeable. The analyses within the levels are uniform and without any trend.

The highest salt concentrations are found in FJ2, 506m and in the first sampled water from FJ4, 420m.

4.4 pH

pH was determined directly in the field and by both laboratories (table 10). The field values are systematically around 5% h gher than the laboratory values. A similar difference has been noted in many groundwaters. There is also a small systematic difference between the laboratories, which may be attributable to the fact that Hydroconsult, which notes the lower value, thermostated its samples at 25° C before measurement.

The differences within the levels are small, as a rule. The highest pH values were measured at the deep levels.

4.5 Conductivity

Conductivity was also measured both in the field and at two laboratories. As a rule, the field value is slightly higher than the laboratory values, but the differences are relatively small. An exception from this rule is FJ4, 420m, where the field value is only about half the laboratory values, which agree closely with each other. The conductivity meter in the field was probably set wrong.

4.6 Organic carbon (TOC)

No parameter exhibits such large differences between the laboratories as TOC - SGAB's results are often more than twice Hydroconsult's results. Control analysis at two Swedish laboratories comes closer to verifying SGAB's high results, while analysis at a laboratory in the United States, which specializes in the determination of humic and fulvic acids, confirmed Hydroconsult's low values. The mean value is reported to only one decimal place (table 8).

The normal concentration lies between 4 and 8 mg/l in FJ2 and FJ4, while it is only about 3 mg/l in FJ8. The levels FJ4, 349m and 420m have very high concentrations at the start of the pumping period, after which the concentrations gradually decline to normal values. Other ions exhibit similar trends at these two levels.

4.7 Negative ions

The negative ions are compiled in tables 7 and 8. The results generally show good consistency within the sampling zones. Where trends occur (4.3), the change is verified by other ions.

Very high nitrogen concentrations, mainly as nitrate, are found in the two middle samples in FJ4, 131m. They have no counterpart in any equally pronounced change in any of the cations analyzed. Agreement between the laboratories' N analyses is good for these two samples. pH and conductivity are determined in the same sample bottle. Normal values are reported by both laboratories.

4.8 Positive ions

The positive ions are compiled in table 6. The concentrations generally differ very little between water samples from the same level. Where trends do exist (4.3), they are verified by changes in several ion concentrations.

4.8.1 Lanthanum, neodymium and ytterbium were determined in the same samples as strontium. In all cases, the concentrations were below the detection limit (< 0.005 mg/l).</p>

5 PARTICULATE MATTER

The particulate matter was subjected to several different types of analyses (5.1-5.5).

5.1 Chemical composition

The water was filtered through a membrane filter (Millipore, pore diameter 0.45 μ m) installed directly after the test chamber. The filtered volume was between 300 and 2300 ml. The total concentration on the membrane was determined by X-ray fluorescence at SGU's (the Geological Survey of Sweden) laboratory in Luleå. The concentrations are presented in table 9 as mg/l of filtered sample. The s. ple volume is also given in the table. The quantity of copper, strontium, lanthanum, neodymium and ytterbium was, with the exception of samples Nos. 260 and 282 (where 1 and 2 μ g Sr, respectively, was obtained), below the detection limit (1, 1, 3, 2 and 2 μ g respectively).

There is a clear correlation between high volume and low concentration, which is natural since the amount of filtrate that can pass a filter is normally dependent on the amount of particles that stay on the filter. The iron and sulphur concentrations also covary, which is not unexpected either.

5.2 Turbidity

Turbidity (table 8) was determined for all samples except those from FJ2, 106m and FJ4, 420m.

5.3 Particle distribution

The particle distribution within the range 2-80 μ m was determined at all levels. The determination was performed by VIAK's water pollution research laboratory in Stockholm. NOTE! It is the length of the borehole and not its vertical depth that is indicated on the curves.

5.4 Humic and fulvic acids

The two deepest levels in the boreholes were analyzed. The determination was performed at Batelle, USA, through combined dialysis and gel film chromotography (GFC). The method, which fractionates the sample into humic acids and four molecular weight fractions of fulvic acid, is described by Means et al 1977 (Limnol. Oceanogr., 22, 957-965). Sodium tetraborate (pH 9.1) was used on recommendation by Swift and Posner 1971 (J. Soil Science, 22, 237-249).

Owing to the samples' low content of organic matter, they were first concentrated ten times before analysis. No loss of volatile organic matter was found in connection with the concentration process.

Level	No.	>1000	700-100	300-700	<300
FJ2, 409m	260	28	63	8	1
FJ2, 506m	266	18	63	15	4
FJ4, 349m	229	48	47	5	0
FJ4, 420m	234	53	37	9	1
FJ8, 402m	285	12	67	18	3
FJ8, 562m	292	23	46	1.7	14

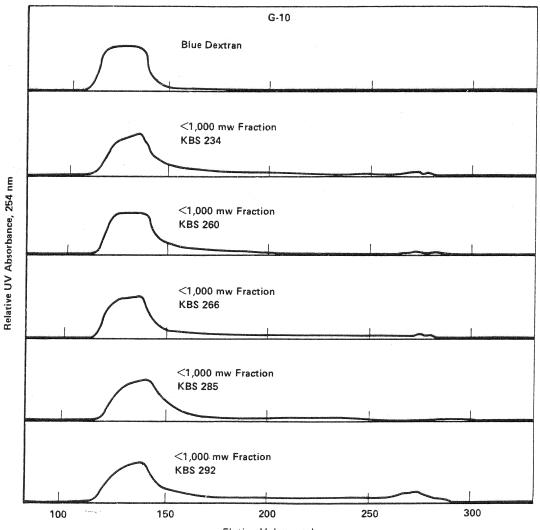
No humic acids were present in the samples. The samples' contents of different molecular weight fractions of fulvic-acid-like substances are shown in the table below. The eluates' UV absorbance, recorded continuously at 254 nm, is shown on the next page. According to Means, the results strongly indicate that the most important organic constituents are hydrophilic polymer substances with low to medium molecular weights - probably fulvic acids or similar substances, which are the predominant substances in natural water.

ISOTOPE ANALYSIS

6

Isotope analysis of light substances in the water is done mainly to permit calculations of the age and origin of the water. The heavy radioactive substances are determined primarily because the natural background concentrations are of great interest for the KBS project.

The analyses were carried out at the following laboratories: Laboratory for Isotope Geology, Stockholm (6.1, 6.3) Institute for Radiohydrometry, Neuherberg, West Germany (6.2, 6.4) Mass Spectrometry Laboratory, Uppsala University (6.4) Studsvik Energiteknik, Nyköping, Sweden (6.5)



6.1 Carbon isotopes

Prior to the determination that is used for dating by means of the carbon-14 method, the water's contents of carbon dioxide and carbonate have to be concentrated. It is thereby of the utmost importance that the concentrate not be contaminated by chemicals or contact with air.

6.1.1 Sample preparation is done in the field. A polyethylene barrel holding 130 l and filled from the beginning with nitrogen is filled with water. Hydrochloric acid is added to disintegrate carbonates present in the water. A nitrogen gas stream is then used to drive the carbon dioxide over to a wash bottle containing sodium hydroxide.

6.1.2 Groundwater age is presented in the table below as "Age BP" and age after correction for C-13 content as "Age BP, corr".

Bore- hole	Depth m	Nr	Age BP	Age BP corr	C13 0/00
FJ2	106	252	4635	4725	-19.5
FJ2	293	256	11010	10960	-28.0
FJ2	409	260	4145	4235	-19.3
FJ2	506	267	13665	13920	- 9.5
FJ4	131	215	3795	3950	-15.3
FJ4	272	222	3345	3475	-17.0
FJ4	349	229	5375	5535	-15.0
FJ4	420	234	6660	6850	-13.3
FJB	402	285	3880	3980	-18.7
FJB	562	292	3890	3975	-19.9

6.2 Deuterium and oxygen

The analyses can furnish information on the climatic conditions prevailing at the time of infiltration.

Bore- hole	Depth m	Nr	H-2 (D)	0-1(3 (D)	9-18 (S)
FJ2	106	252	-80.5	-11.33	11 24
FJ2	293	256			-11 31
FJ2	409	260			-11, 35
FJ2	506	267			-14.11
FJ4	1:31	216			-11.50
FJ4	272	222	-82.6	-11.54	11.57
FJ4	349	229	81.6	-11.45	-11.73
FJ4	420	234	-84.7	-11.69	t t . 88
FJB	402	285	-79.3	-11.22	11, 21
FJB	562	292	-77.8	-10.94	-11.16

The concentrations in the table refer to deviations in per mill from SMOW (Standard Mean Oceanic Water). The analyses were performed in West Germany (D) and in Sweden (S).

6.3 Tritium

Owing to its short half-life (about 12 years), tritium is an important isotope in hydrologic studies. The amount of tritium in the atmosphere has increased drastically due to nuclear weapons testing. The tritium content of rain water has increased by more than ten times, which makes it possible to determine whether "young" water is present in a groundwater sample.

The tritium concentration is given in the tables in the unit TU, which is the number of tritium atoms per 10^{18} hydrogen atoms.

No tritium determination has been carried out on the drilling water used here.

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Nr	Bore-	Depth	Tr
	hole	m	TU
1			
256	FJ2	293	19
260	FJ2	409	19
267	FJ2	506	<3
215	FJ4	131	9
222	FJ4	272	21
229	FJ4	349	12
234	FJ4	420	6
285	FJ8	402	8
	FJB	562	10
Ross / Loss	8 3. See 9	Inf and Seno	

6.4 Uranium, thorium, radium and radon

The natural concentration of these elements in the groundwater is of great interest for the project. The analyses are presented in table 10. The concentrations for U, Ra and Rn are given in Bq/1. The following relationships apply:

39.4 μ g U per Bq 2.74 x 10⁻⁵ μ g Ra-226 per Bq 1.76 x 10⁻¹⁰ μ g Rn-222 per Bq

GAS ANALYSIS

7

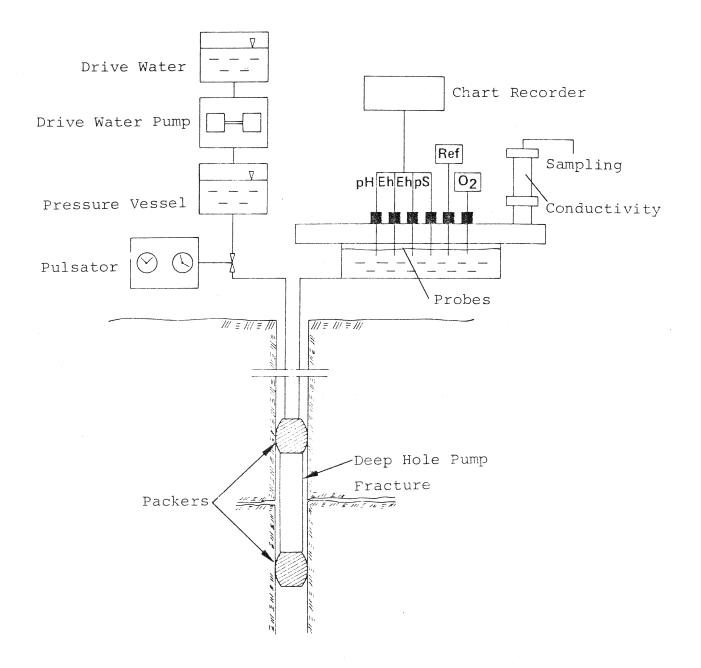
Helium was determined in the gas that spontaneously leaves the water on its passage through a Horst bottle. The analysis was carried out by AGA SpecialGas, Lidingö, Sweden. No gas sample was taken from FJ4, 349m.

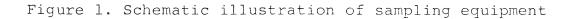
In the table below, besides the percentage content of helium in the gas phase, the amount of water that has passed through the Horst bottle and the obtained gas volume are also given. This enables the reader to form his own rough idea of the quantity of dissolved gases in the water. The helium content of the water has not been calculated, since degassing is dependent on a number of uncontrolled parameters. We know, for example, that temperature variations of >20 °C occur in the tent where the borehole opens out and the sampling equipment is located.

BH	Depth m	No.	Helium %	Water volume 1	Gas volume m ³	Notes
FJ2	106	252	0.099	218	90	
	293	256	0.002	217	120	
	409	260	0.008	103	145	
	506	267	1.1	19	190	
FJ4	131	215		122	185	2
	272	222	0.004	240		1
	420	234	0.005	195		1
FJ8	402	285	0.003	185	90	
	562	292	0.004	84		1

Note 1 Data on gas volume lacking.

Note 2 Gas analysis lacking.





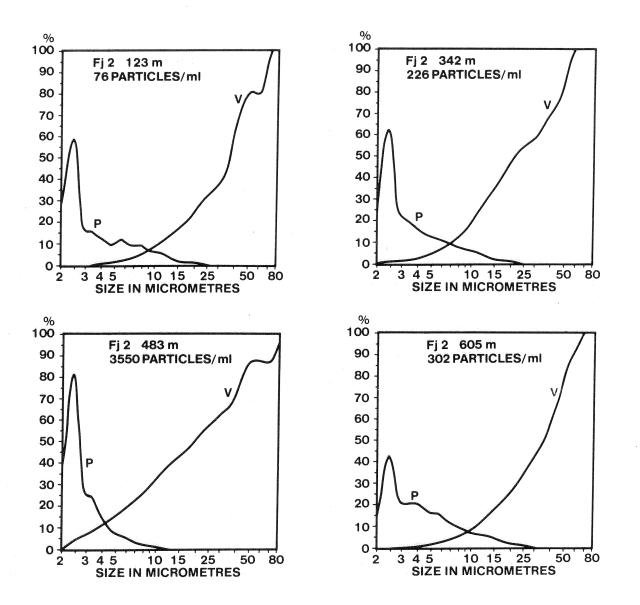


Figure 2a. Particle size distribution analysis in the region $2-80~\mu$ m. An HIAC PA 500 instrument is used. P represents the percent of the total number of particles with a certain particle size. V represents the percent of the total numer particle volume less than a certain particle size.

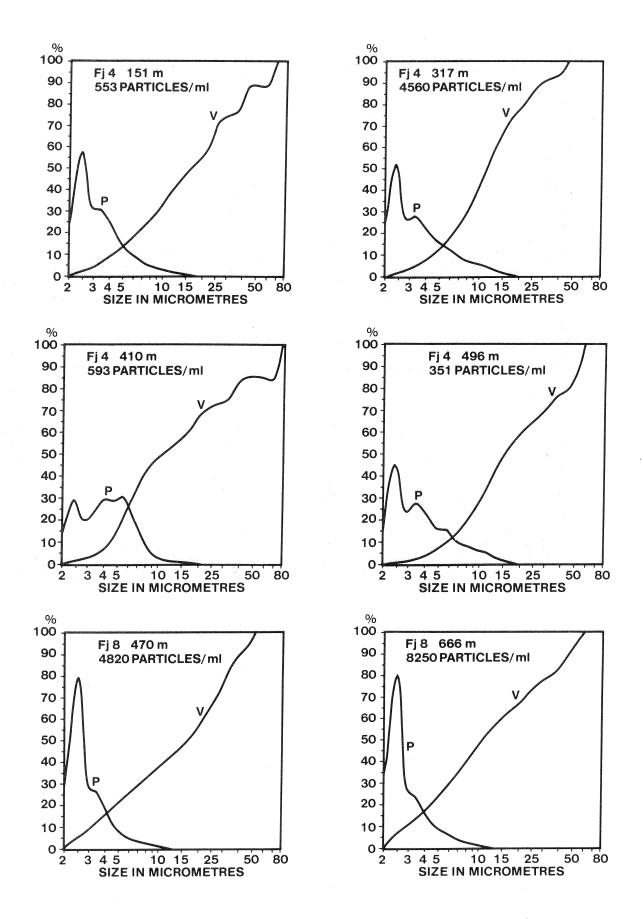


Figure 2b. Particle size distribution analysis in the region 2-80 μ m. An HIAC PA 500 instrument is used. P represents the percent of the total number of particles with a certain particle size. V represents the percent of the total particle volume less than a certain particle size.

FJALLVEDEN - Field measurements

Table 1

Cali- bration nr	Bore- hole	Hole- length m	Date ww-d	Time	Conduc- tivity mS/m	Oxy− gen mg/l	pН	Eh,C mV	Eh,Pt mV	βĘ
301	FJ2	123	34-6 34-6 34-6 34-6 34-7 34-7 34-7 34-7 34-7 34-7	830 1030 1230 1530 745 930 1130 1230 1330	26. 9 26. 9 26. 9 26. 9 26. 9 27. 0 27. 0 27. 0 27. 0 27. 0 27. 0	. 00 . 00 . 00 . 00 . 00 . 00 . 00 . 00	8.0 8.0 8.0 8.0 8.0 8.2 8.1 8.1 8.2 8.1	- 59 - 62 - 65 - 66 - 88 - 88 - 88 - 88 - 88 - 88 - 88	4 1 - 2 - 3 - 5 - 16 - 18 - 20 - 20 - 11	13.0 12.6 12.3 12.9 12.0 11.2 11.1 11.0 11.1
304	FJ2	342	36-4 36-5 36-5	1600 730 900	27.5 27.1 27.3	. 04 . 02 . 00	7.1 7.2 7.2	- 4 - 17 - 20	81 80 76	20.0 18.5 18.2
305	FJ2	342	36-7 36-7 36-7 36-7 36-7	700 900 1140 1330 1450	27.0 27.3 27.4 27.4 27.4 27.4		7.2 7.1 7.1 7.1 7.1 7.1	- 22 - 27 - 31 - 36 - 38	71 63 58 51 50	16.9 16.3 15.5 15.1 14.8
306	FJ2	342	37-2 37-2	650 730	27.5 27.3	. 05 . 10	7. 3 7. 3	- 42 - 43	87 87	21.3 21.3
308	FJ2	483	38-6 38-6 38-6 38-6 38-7 38-7 38-7 38-7 38-7	700 900 1130 1240 1500 645 845 1100 1200	29. 2 29. 2 29. 4 29. 5 29. 5 30. 0 30. 0 30. 0 30. 0	. 02 . 02 . 02 . 02 . 02 . 02 . 02 . 02	7.55 7.55 7.55 7.55 7.55 7.55 7.55	- 92 - 95 -103 -104 -116 -115 -119 -118 -119	14 11 14 13 - 13 - 15 - 18 - 19	17.7 17.5 17.4 17.4 17.2 15.7 15.6 15.4 15.3
309	FJ2	483	39-2 39-2 39-2	545 645 750	30. 9 30. 9 30. 9	. 00 . 00 . 00	7.5 7.5 7.5	- 87 - 88 - 89	19 18 17	17.4 17.3 17.2
311	FJ2	605	40-3 40-3 40-3 40-3 40-4 40-4 40-4 40-4	715 920 1145 1305 1510 800 1015 1215 1430 1545 735 905	71.1 71.7 71.5 71.8 72.0 72.9 73.5 73.4 73.7 73.7 73.9 74.9 74.9	. 00 . 03 . 03 . 02 . 02 . 02 . 00 . 03 . 01 . 01 . 02 . 01	8.88 8.88 8.89 8.99 8.89 8.88 8.88 8.88	-184 -188 -101 -109 -105 -105 -103 -107 -111 -109 -109 -110	-119 -121 -123 -123 -123 -122 -117 -124 -125 -126 -129 -128	13.0 12.7 12.5 12.3 12.1 11.1 11.0 10.9 10.8 10.7 10.4 10.4
312	FJ2	605	40-7 40-7 40-7 40-7 40-7	710 900 1025 1140 1250	76.5 76.4 77.0 77.2 77.2	. 10 . 10 . 10 . 10 . 20	8.9 8.9 8.9 8.9 8.9 8.9	- 29 - 29 - 33 - 35 - 17	- 21 - 20 - 25 - 27 - 11	19.8 20.0 19.6 19.5 20.0
313	FJ2	605	41-2 41-2 41-2	755 1000 1100	78.9 79.0 78.9	. 10 . 10 . 10	8.9 8.9 8.9	- 13 - 20 - 21	- 12 - 20 - 23	19.9 19.6 19.5

Table 2

FJALLVEDEN - Field measurements

Cali- bration nr	Bore- hole	Hole- length m	Date ww-d	Time	Conduc− tivity mS/m	Oxy− gen mg/l	рН	Eh,C mV	Eh,Pt mV	рS
202	FJ4	151	26-3 26-3 26-3 26-3 26-3 26-4 26-4 26-4 26-4 26-4 26-4 26-4 26-4	730 945 1140 1230 1335 1435 730 834 1035 1225 1300 1410 1510	$\begin{array}{cccccccccccccccccccccccccccccccccccc$		7.9 7.8 7.8 7.9 7.9 7.9 7.9 7.7 7.7 7.8	- 6 2 1 2 11 40 21 35 34 34 34 34 34 35	32 41 42 44 48 113 118 120 116 113 107 108	19.9 20.1 20.0 19.9 20.1 21.8 21.8 21.8 21.7 21.4 21.7 21.8 21.8
203	FJ4	151	26-6 26-6 26-7 26-7 26-7	1200 1420 1600 735 830	33.4 33.4 33.0 32.9 33.0	. 00 . 00 . 00 . 00 . 00		24 21 15 29 28	54 50 43 61 63	19.2 19.1 19.0 19.0 19.0
204	FJ4	151	27-1 27-1 27-1 27-1 27-1 27-1 27-2	1300 1430 1600 1700 1800 1200	35.0 35.1 35.2 35.3 35.4 35.3	. 04 . 05 . 04 . 04 . 04 . 04		- 83 - 87 - 85 - 83 - 88 - 82	- 62 - 64 - 61 - 59 - 62 - 57	13.2 12.6 12.4 12.4 12.1 12.1
204	FJ4	317	28-3 28-3 28-3 28-3 28-3 28-3 28-3 28-3	645 800 900 1200 1320 1430 1530 650	33. 4 33. 4 33. 4 33. 4 33. 5 33. 5 33. 5 33. 5 33. 4	00 00 01 01 02 02 02		- 76 - 79 - 81 - 84 - 85 - 86 - 88 - 125	- 17 - 18 - 19 - 23 - 25 - 26 - 26 - 115	13. 1 13. 0 13. 0 12. 8 12. 7 12. 7 12. 7 12. 6
207	FJ4	317	28-6 28-6 28-6 28-7 28-7 28-7	1225 1400 1500 1635 750 920	33, 1 33, 2 33, 3 33, 3 33, 4 33, 7	. 04 . 04 . 04		- 24 - 25 - 21 - 20 - 28 - 26	- 16 - 14 - 11 - 12 - 19 - 17	19.4 18.8 18.9 18.9 18.0 17.8
208	FJ4	317	29-2 29-2 29-2 29-2	645 815 845 1000	34. 0 34. 0 34. 0 33. 9	. 00 . 01 . 02 . 03		- 75 - 83 - 85 - 90	3 1 1	14.6 14.5 14.5 14.4
209	FJ4	410	30-1 30-1	835 910	38. 1 38. 1	. 00 . 00		-178 -178	- 17 - 17	11.6 11.6
211	FJ4	410	30-6 30-6 30-6 30-6 30-6 30-7 30-7 30-7	635 800 1210 1400 800 900 930	31. 2 31. 2 31. 1 31. 2 31. 1 31. 5 31. 5 31. 5	. 00 . 03 . 05 . 05 . 05 . 00 . 00 . 00		-115 -118 -122 -130 -130 -156 -161 - 75	- 64 - 67 - 70 - 76 - 77 - 85 - 85 - 87	11.4 11.3 11.3 11.2 11.2 11.3 11.3 11.3
212	FJ4	410	31-1 31-2 31-2 31-2 31-2 31-2 31-2 31-2	1405 1530 745 745 840 930 1030 1130	32.0 32.0 31.8 31.8 31.9 31.9 31.9 31.9 32.0	. 00 . 00 . 00 . 00 . 00 . 01 . 03 . 04		-100 - 99 -109 -109 -109 -109 -109 -110 -109	- 89 - 89 - 84 - 84 - 84 - 84 - 85 - 86 - 87	13. 1 13. 0 12. 5 12. 5 12. 5 12. 5 12. 5 12. 4 12. 3

FJALLVEDEN - Field measurements

Table 3

Cali- bration nr	Bore- hole	Hole- length m	Date ww-d	Time	Conduc− tivity mS/m	Oxy- gen mg∕l	pH	Eh,C m∀	Eh,C ex mV	Eh,Pt mV	Eh,Pt ex mV	рS
214	FJ4	496	34-6 34-6 34-6 34-6 34-6 34-6 34-7 34-7	735 930 1145 1400 1500 1600 740 900	20.0 20.0 19.9 19.6 19.5 19.5 18.1 18.2			-171 -173 -173 -174 -174 -174 -194 -195	- 28 - 39 - 36 -219 -220 -221 -231 -233	-140 -144 -149 -156 -157 -158 -163 -165	- 43 - 39 - 47 -227 -228 -228 -235 -237	11.5 11.4 11.2 11.2 11.2 10.9 10.9
215	FJ4	496	35-1 35-1 35-1 35-2 35-2 35-2 35-2 35-2 35-2	1130 1340 1400 1740 800 955 1100 1245 1330	17.5 17.5 17.4 17.5 17.5 17.5 17.5 17.5 17.5	. 03 . 05 . 05 . 02 . 00 . 00 . 00 . 02 . 02 . 02		-160 -164 -165 -166 -167 -171 -171 -173 -173	-122 -128 -128 -129 -134 -140 -141 -141 -145	-154 -158 -159 -159 -159 -162 -163 -163 -164 -165	-142 -148 -155 -160 - 70 -102 -110 -117 -123	12.6 12.4 12.2 12.2 11.7 11.7 11.7 11.6 11.6
402	FJB	470	38-7 38-7 38-7 38-7	730 850 930 1205	23.8 23.5 23.8 23.8 23.8	. 01 . 01 . 03 . 02	8.4 8.4 8.4 8.4	-109 -109 -109 -110	- 65 - 67 - 66 - 75	-102 -103 -103 -103	- 82 - 85 - 85 - 94	16.4 16.3 16.3 16.3
403	FJB	470	39-1 39-2 39-2 39-2 39-2 39-2 39-2 39-2	1510 1630 745 930 1030 1215 1440	23. 9 23. 9 23. 9 23. 9 23. 9 23. 9 23. 9 23. 9	01 01 00 01 01 01	8.6 8.5 8.5 8.5 8.5 8.5 8.5	-118 -119 -119 -121 -121 -123 -124	- 86 - 87 - 96 - 99 -101 -105 -108	-113 -113 -113 -115 -116 -117 -119	-114 -116 -133 -137 -139 -143 -143	13.9 13.9 13.6 13.5 13.5 13.4 13.3
405	FJB	666	40-2 40-3 40-3 40-3 40-3 40-3 40-4 40-4 40-4	1230 1530 750 930 1100 1315 1530 750 1000 1200 1400 1550 750 900 1405	24. 0 24. 0	. 01 . 01 . 01 . 02 . 01 . 02 . 01 . 00 . 00 . 00 . 00 . 00 . 00 . 00	8.99 9.09 8.99 9.00 9.00 9.00 9.00 9.1 1 9.1	-153 -154 -155 -157 -158 -159 -169 -169 -171 -173 -175 -177 -198 -199 -154	-156 -159 -176 -178 -181 -184 -201 -204 -205 -206 -207 -211 -212 -158	-140 -142 -141 -142 -144 -145 -144 -140 -141 -143 -143 -143 -143 -143 -143 -141 -141	-174 -180 -203 -205 -208 -211 -214 -228 -228 -229 -230 -235 -235 -177	12.7 12.4 12.4 12.3 12.3 12.3 12.3 12.2 12.1 12.1 12.1
406	FJB	666	40-6 40-6 40-6 40-7 40-7	1140 1400 1525 740 845	24. 0 24. 0 24. 0 24. 0 24. 0 24. 0	. 00 . 00 . 00 . 00 . 00	9.0 9.0 9.0 9.0 9.0 9.0	-136 -138 -138 -142 -142	-145 -150 -151 -165 -165	-128 -129 -129 -131 -131	-161 -167 -170 -192 -190	13.0 12.9 12.9 12.5 12.4
407	FJB	666	41-1 41-1 41-2 41-2 41-2 41-2 41-3 41-3	1300 1500 1600 750 935 1120 1430 800 915	24. 0 24. 0 24. 0 24. 0 24. 0 24. 0 24. 0 24. 1 24. 1	. 03 . 02 . 02 . 02 . 02 . 03 . 02 . 03 . 02 . 01 . 04	8.9 8.9 8.9 8.9 8.9 8.9 8.9 8.9 8.9 8.9	-141 -141 -143 -143 -144 -143 -144 -143 -141	-168 -169 -162 -184 -186 -188 -190 -199 -201	-132 -132 -132 -132 -132 -133 -134 -129 -130	-188 -189 -184 -207 -209 -211 -213 -223 -224	12.4 12.3 12.1 12.1 12.1 12.1 12.0 12.0 12.0

FJÄLLVEDEN - E⁰-values

Table 4

Bore- hole	Hole- length m	Cali- bration nr	Date ww-d	Glass pH4 mV	Glass pH7 mV	Glass pHiO mV	C pH4 mV	C pH7 mV	₽t pH4 mV	Pt pH7 mV	Ag,S O.O1M mV	Ag, S 0.05M mV
FJ2	123	301 302	34-4 34-7	391 385	396 388	400 395	283 259	259 257	278 256	256 257	-841 -845	-835 -835
FJ2	342	304 305 306	36-2 36-5 36-7	384 380 390	384 378 392	382 383 399	258 262 252	263 265 253	253 259 251	256 261 250	-822 -819 -829	-824 -814 -822
FJ2	483	307 308 309 310	37-2 38-4 38-7 39-2	382 379 378 484	384 383 376 482	391 387 407 390	263 258 265 264	255 257 259 259	253 255 256 258	248 255 259 256	-820 -809 -823 -817	-810 -805 -813 -809
FJ2	605	311 312 313 314	40-1 40-5 40-7 41-2	385 381 381 377	384 382 381 377	374 383 380 376	251 254 263 278	257 251 260 267	249 254 254 257	254 254 255 259	-820 -824 -827 -844	-823 -827 -832 -836
FJ4	151	201 202 203 204	25-2 26-1 26-5 26-7	392 389 383 453	395 383 381 470	389 383 397 483	250 251 252 259	249 253 254 257	246 244 251 255	247 248 253 256	-859 -836 -821 -809	-852 -827 -824 -816
FJ4	317	205 206 207 208	27-3 28-1 28-4 28-7	375 438 542 605	389 463 563 630	364 481 535 645	254 256 257 246	253 248 258 246	254 255 256 242	252 252 256 243	-827 -838 -835 -807	-824 -826 -833 -816
FJ4	410	209 210 211 212	29-3 30-1 30-4 30-7	563 806 342 233	592 761 528 409	610 810 575 558	250 244 251 251	251 244 237 240	250 243 249 250	251 244 243 240	-835 -839 -845 -868	-822 -830 -825 -854
FJ4	496	213 214 215 216	31-4 34-4 34-7 35-2	629 434 554 562	665 437 578 592	682 448 604 619	245 225 256 260	245 252 252 249	244 229 258 267	243 252 252 250	-878 -825 -841 -150	-857 -807 -824 - 47
FJB	470	401 402 403	38-2 38-5 38-7	394 384 388	413 386 390	400 388 392	253 264 251	282 246 250	253 269 247	282 244 247	-895 -838 -840	-873 -846 -846
FJ8	666	404 405 406 407 408	39-3 40-1 40-5 40-7 41-3	390 393 60 391 382	396 402 404 402 409	401 405 731 408 394	253 255 259 258 258	252 250 256 254 255	250 252 258 254 253	250 247 256 254 254	-836 -853 -836 -838 -831	-840 -841 -829 -829 -831

FJALLVEDEN - Field	measurements	and	corresponding	laboratory	values
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Table 5a

Table 5b

Bore- hole	Depth	Hole- length m	Νr	Date sampling yy-ww-d	Date lab ww-d	pH field	pH 1≈b	Eh,C field mV	EH,₽t field mV	pS field	52- lab mq∕l	Cond. field mS∕m	Cond. lab mS∕m	Oxygen field mg∕l
				39-										2.1
FJ2	106	123												
			239	82-34-4	34-5	7.7	6.7	70	55	22.3	. 05	26	23.	. 30
			240	82-34-6	35-2	8.0	7.7	- 66	- 3	12.2	. 08	27	27	<. 02
			251	82-34-7	35-2	8.1	7.7	· 88	- 10	11.1	. 06	27	27	< 02
			252	82-35-1	35-2	8.0	7.7	2	15	19.2	. 06	27	27	. 05
FJ2	293	342												
102	E73	342	253	82-34-2	36-3		7.1				. 03		25	<. 02
			254	82-36-3	36-4	7.1	6.9	29	44	22.8	. 03	27	25	. 30
			255	82-36-4	36-5	7.1	6.9	2	81	20.6	. 04	28	25	. 00
			256	82-37-1	37-2	7.3	7.0	- 20	92	22.5	. 08	27	26	. 20
FJ2	409	483	~ ~ 7			7 0		05	00	23. 1	0.1	~ 7	05	70
			257	82-38-4	38-5	7.2	7.2	85	80 -209	17.4	.01	27 30	25 28	. 70
			258	82-38-6	39-2	7.5 7.5	7.3	-140	- 18	15.4	. 03	30	28	. 02 -
			259	82-38-7 82-39-1	39-2	7.3	7.4	- 40	43	20.1	. 02	30	20	. 02
			260	82-39-1	39-2		7.4	- 88	18	17.3	. 02	30 31	28	<. 02
			261	85-34-5	39-4	7.5	7.4	- 68	10	17.3	. 03	- 1	20	N. 02
FJ2	506	605												
			262	82-40-1	40-2	8.6	ፀ. ሪ	- 55	9	-22.3	. 02	66	63	. 40
			263	82-40-2	40-3	8.7	3.6	-125	- 25	20. 9	. 04	69	65	. 20
			264	82-40-3	40-4	8.8	9.6	-105	-123	12,4	. 11	72	68	. 03
			265	82-40-4	40-5	8.8	B. 7	1 07	-124	10.9	. 11	73	71	. 03
			266	82-40-7	41-2	8.9	8.8	- 34	- 26	19.5	. 01	77	73	. 10
			267	82-41-1	41-2	8.9	8.8	25	35	21.3	. 01	78	74	. 30
			268	82-41-2	41-4	8.9	9.7	- 20	- 20	19.6	< 01	79	72	. 10
			905				8.1						30	
			907				8.2						29	

FJALLVEDEN -	Field	measurements	and	corresponding	laboratoru	values

Bore- hole	Depth	Hole- length m	Nr	Date sampling yy-ww-d	Date lab ww-d	pH field	pH lab	Eh,C field mV	EH,Pt field mV	p5 field	S2- lab mg∕l	Cond. field mS∕m	Cond. lab mS∕m	Oxygen field mg∕l
4لF	131	151	211 212 213 214 215 216	82-26-2 82-26-3 82-26-4 82-26-7 82-27-1 82-27-1	26-3 26-4 26-5 27-2 27-2 27-3	7.9 7.8 7.7	7.7 7.7 7.5 7.5 8.2 7.8	2 1 36 - 83 - 34	22 42 107 - 62 - 8	20.8 19.9 21.8 13.2 14.7	. 04 . 04 . 02 . 05 . 05 . 05	34 35 33 35 35 34	33 34 32 35 34	. 05 . 04 . 03
FJ4	272	317	217 218 219 220 221 222 223	82281 82282 82283 82284 82287 82291 82292	28-2 29-3 28-4 28-5 29-2 29-2 29-3		7.6 7.5 7.5 7.5 7.5 7.5 7.4	- 47 - 86 - 90	- 23 - 26	13.6 12.7 15.2 14.4	03 02 04 02 < 01 01 03	33 33 34 33 34 34 34 34	30 32 32 33 33 33 33	. 10 . 03 . 02 . 04 . 02
¥ن¥	349	410	224 225 226 227 228 229 230	82-30-1 82-30-2 82-30-3 82-30-4 82-30-7 82-31-1 82-31-2	30-2 30-3 30-4 30-5 31-2 31-2 31-3		7, 6 7, 6 8, 1 9, 2 9, 3 9, 3 9, 3 9, 0	41 - 29 - 73 - 85 - 99 -110	49 - 18 - 54 -101 - 95 - 86	21.5 18.9 12.8 11.3 17.9 13.3 12.4	. 05 . 13 . 14 . 20 . 05 . 04 . 04	37 37 36 35 32 32 32	37 36 35 32 32 32	.50 .20 .10 .06 .03 <.02 .03
F ئ4	420	496	231 232 233 234 235 904 904	82-34-4 82-34-6 82-34-7 82-35-1 82-35-2	34-5 35-2 35-2 35-2 35-3		6.8 7.9 8.3 8.4 3.5 (3.1 (3.1	-174 -125 -160 -172	-156 -120 -154 -164	14.0 11.2 14.4 12.6 11.6	.05 .13 .13 .13 .13 .01	32 20 18 18 18	62 37 34 33 33 29 30	. 08 . 02 . 07 . 03 . 02

FJALLVEDEN - Field measurements and corresponding laboratory values

Table 5c

Bore- hole	Depth m	Hole- length m	Nr	Date sampling yy-ww-d	Date lab ww-d	pH field	pH lab	Eh,C field mV	EH,Pt field mV	pS field	52- lab mg/l	Cond. field mS∕m	Cond. lab mS∕m	Oxygen field mg/l
FJ8	402	470	281 282 283 284 285 286	82-38-3 82-38-4 82-38-5 82-38-7 82-39-1 82-39-2	38-4 38-5 39-2 39-3 39-3 39-4	8.0 8.3 8.4 8.4 8.5	7.6 7.7 7.9 7.9 8.0 7.9	- 53 - 5 - 97 -109 -117 -121	- 45 - 10 - 71 -103 -111 -116	18.2 21.0 16.8 16.3 14.0 13.5	. 05 . 03 . 01 . 01 . 02	23 23 24 24 24 24	22 22 22 22 22 22 22 22	04 20 04 03 02 < 02
FJ8	562	666	287 288 289 290 291 292 293	82-40-1 82-40-2 82-40-3 82-40-4 82-40-7 82-41-1 82-41-2	40-2 40-3 40-4 40-5 41-2 41-2 41-2	8.6 8.9 9.0 8.7 8.7 8.9 8.9	7.9 7.9 7.9 7.9 7.9 7.9 7.9 7.8	-134 -153 -159 -174 -106 -139 -144	-127 -140 -145 -143 -108 -130 -133	13.7 12.7 12.3 12.1 14.0 12.5 12.1	. 02 . 09 . 08 . 08 . 01 . 01 <. 01	24 24 24 24 24 24 24 24	22 22 22 22 22 22 22 22 22 22	02 < 02 < 02 < 02 04 03 03
			908 909				7.3 7.5						24 23	

Table 6a

FJÄLLVEDEN - Metal ions

Bore- hole	Depth m	Nr	Ca mg∕l	Mg mg∕l	Na mg∕l	K mg∕l	Mn mg∕l	Fe⊋+ mg/l	Fe-tot mg∕l	Al mg∕l	Cu mg∕l	Sr mg∕l
FJ2	106											
		239	23	3.9	27	2.4	. 33	1.6	2.0	. 03	<. 005	. 124
		240	22	3.9	35	2.3	. 22	1.0	1.1	. 02	<. 005	. 128
		251	18	3.2	37	2.4	. 18	0.t	1.1	<. 01	<. 005	. 180
		252	18	3.0	37	2.4	. 18	. 95	1.1	<. 01	<. 005	. 125
FJ2	293											
		253	20	3.2	26	2.3	. 54	5.1	12			
		254	19	3.3	27	2.3	. 49	6.2	10			
		255	18	3.3	26	2.3	. 56	5.7	12			. 125
		256	19	3.3	26	2.3	. 49					. 122
FJ2	409											
		257	21	3.8	28	2.4	. 48	8.0	9.9	< 01	<. 005	. 125
		258	21	3.6	31	2.5	. 45	7.0	7.1			
		259	22	3.6	32	2.6	. 40	6.3	6.4			
		260	21	3.4	33	2.6	. 46	6.5	7.0	< 01	. 005	. 125
		261	21	3.6	34	2.6	. 40	4.7	5.5			
FJ2	506											
		262	13	1.2	113	1.2	. 08	. 42	. 83			
		263	13	1.2	117	1.2	. 08	. 32	. 83			
		264	13	1.3	122	1.2	. 07	. 42	. 52			
		265	13	1.0	125	1.0	. 07	. 40	. 48	<. 01	<. 005	. 125
		266	12	. 9	130	1.0	. 06	. 24	. 33			
		267	12	. 8	130	1.0	. 06	. 22	. 34	< 01	< 005	. 133
		268	12	1.2	129	1.0	. 06	. 42	. 42			
		905	31	5.0	26	4.7	. 06		. 10			
		907	13	2.9	49	2.6	. 10		. 27			

FUXLLVEDEN - Metal ions

Table 6b

$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	Al Cu Sr ng/l mg/l mg/l	Fe-tot Al mg/l mg/l	Fe2+ mg/l	Mrı mg∕l	K mg∕l	Na mg∕ì	Mg mg∕l	Ca mg∕l	Nr	Depth m	Bore- hole
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$										131	FJ4
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$											
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	10 . 005 . 100										
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$.10 <.005 .192										
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	10 (005 145										
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$.10 <.005 .145										
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$		2.0	1.6	. 10	3.0	62	2.5	15	216		
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$										272	F.J4
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$		7.0	6.1	. 24	2.8	38	3.6	28	217		
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$		7.3									
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$		7.5	6.5								
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$.05 <.005 .212	7.3 .05	6.4								
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$		7.8	6.4								
F.J4 349 F.J4 349 224 15 1.5 72 2.0 .10 1.6 2.7 225 13 1.3 72 1.8 .07 1.3 2.3 226 11 .9 71 1.7 .04 .35 .70 227 10 .8 71 1.6 .04 .21 .45 .14 < 005 228 17 2.4 54 2.3 .07 \cdot .99 1.0 229 17 2.3 54 2.4 .09 1.37 1.4 < 01 < 005 230 17 2.3 56 2.5 .09 J.78 1.8 F.J4 420 F.J4 420 F.J4 420 231 21 2.0 100 2.8 .98 3.3 11.0 .07 < .005 232 13 1.8 68 2.1 .11 2.3 2.3 .01 < .005 233 13 1.8 62 2.0 .10 1.6 1.6 < .01 < .005 234 14 2.0 62 2.0 .09 1.2 1.4 < .01 < .005	.14 < 005 .168	10 . 14	8.1	. 30							
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$		9.8	8.2	. 29	2.7	37	3.9	29	223		
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$										240	F 1.4
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$		2 7	1 4	10	2.0	70		15	224	347	F J4
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$											
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$											
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$. 14 <. 005 . 092										
Z29 17 2.3 54 2.4 .09 1.37 1.4 < 01 < 005 Z30 17 2.3 56 2.5 .09 1.78 1.8											
FJ4 420 231 21 2.0 100 2.8 .28 3.3 11.0 .07 <.005	< 01 < 005 142										
FJ4 420 231 21 2.0 100 2.8 .28 3.3 11.0 .07 <.005 232 13 1.8 68 2.1 .11 2.3 2.3 .01 <.005 233 13 1.8 62 2.0 .10 1.6 1.6 <.01 <.005 234 14 2.0 62 2.0 .09 1.2 1.4 <.01 <.005											
231 21 2.0 100 2.8 28 3.3 11.0 .07 <.005							Rend C Col	27			
232 13 1.8 68 2.1 .11 2.3 2.3 .01 <.005										420	FJ4
233 13 1.8 62 2.0 .10 1.6 1.6 <.01											
234 14 2.0 62 2.0 .07 1.2 1.4 < 01 < 005											
235 14 2.2 62 2.0 .07 1.3 1.3 <.01 <.005	4.01 <.005 .115	1.3 <.01	1.3	. 09	2.0	62	2.2	14	235		
904 13 2.7 50 2.7 JO .90											
906 32 4.8 28 4.7 .06 .07		. 07		. 06	4.7	28	4.8	32	906		

										т	able 6c	
				F	JALLVEDE	EN - Meta	al ions					
Bore- hole	Depth m	Nr	Ca mg∕l	Mg mg∕l	Na mg∕1	K mg∕l	Mn mg∕l	Fe2+ mg/l	Fe-tot mg/1	Al mg/l	Cu mg∕1	Sr mg⁄1
FJB	402											
		281	25	4.2	13	3.1	. 16	3.5	3. 8			
		282	25	4.4	13	3.1	. 14	2.1	2.5	<. 01	.<. 005	. 091
		283	25	4.6	13	3.2	. 15	2.4	3.0			
		284	25	4.6	13	3.2	. 15					
		285	25	4.6	13	3.2	. 15	2.2	2.9			
		286	25	4.6	13	3.1	. 15	2.4	2. 9			
FJB	562											
		287	26	4.2	13	3.0	. 16	2.8	3.7			
		288	26	4.2	13	3.0	. 16	2.8	3.1			
		289	26	4.2	13	3.0	. 15	2.8	3.1			
		290	26	4.2	13	3.0	. 15	2.7	3. t	< 01	<. 005	. 091
		291	26	4.1	13	3.0	. 15	2.6	2.9			
		292	26	4.0	14	2.7	. 15	2.7	3.2	< 01	< 005	. 093
		293	26	4. 2	13	3.0	. 15	2.5	3.1			
			-	4.5	10	0.7	0E		D/			
		908	32	4.3	10	3.7	35		. 06			
		909	29	4.0	10	3.7	. 33		. 07			

FUALLVEDEN - Nitrogen containing ions

Bore-	Depth	Nr	N02	ND3 mg/1	NH4 mg/l		N02-N M021	ND3-N mg/1	NH4−N ma∕1	Sum-N
hole	m		mg / 1	mgvi	mg / 1		my/i	mg / 1	mg / I	mg∕1
FJ2	106									
		239	. 004	. 045	. 115		. 001	. 010	. 089	. 101
		240	. 005	. 020	. 085		. 002	. 005	. 066	072
		251	. 007	. 015	. 080		. 002	. 003	062	. 068
		252	<. 002	. 030	. 090		<. 001	. 007	. 070	. 077
FJ2	293	2000 - 100 -	the second se	1.5.	1.1			- /		
		253	. 004	. 040	. 070		. 001	. 009	. 054	. 065
		254	<. 002	. 040	. 065		<. 001	. 009	. 050	. 060
		255	< 002	. 025	. 075		<. 001	. 006	. 058	. 064
		256	. 005	. 020	. 055		. 002	. 005	. 043	. 049
FJ2	400									
FUZ	409	057	007	. 030	050		. 002	007	000	0.10
		257	. 007		. 050		<. 001	. 007	. 039	. 048
		258	< 002	. 060	. 050		<. 001	. 014	. 039	. 053
		259	< 002	. 060	. 055	× ·	<. 001	. 014	. 043	. 057
		260	< 002	. 065	. 040		002	. 015	. 031	. 046
		261	. 005	. 035	. 065		. 00e	. 008	. 050	. 060
FJ2	506									
		262	. 004	. 040	. 015		. 001	. 009	. 012	. 022
		263	. 005	. 030	. 010		. 002	. 007	. 008	. 016
		264	<. 002	. 030	. 010		<. 001	. 007	. 008	. 015
		265	<. 002	. 055	. 010		<. 001	. 012	. 008	. 021
		266	< 002	. 020	. 010		<. 001	. 005	. 008	. 013
		267	<. 002	. 030	. 010		<. 001	. 007	. 008	. 015
		268	<. 002	. 045	. 050		<. 001	. 010	. 039	. 050

Table 7b

FJALLVEDEN - Nitrogen containing ions

Bore- hole	Depth m	Nr	NO2 mg/1	ND3 mg/1	NH4 mg∕1	NO2-N mg/l	ND3−N mg / 1	NH4-N mg∕l	Sum-N mg∕1
FJ4	131								
		211	. 005	. 040	. 120	. 002	. 009	. 093	. 104
		212	. 005	. 055	. 110	. 002	. 012	. 085	. 099
		213	. 095	4. 500	. 125	. 029	1.017	. 097	1.143
		214	. 520	1.570	. 230	. 158	. 355	. 179	. 692
		215	. 007	. 060	. 120	. 002	. 014	. 093	. 109
		216	. 027	. 025	. 260	. 008	. 006	. 202	. 216
FJ4	272								
		217	. 015	. 770	. 075	. 005	. 174	. 058	. 237
		218	. 015	. 065	. 135	. 005	. 015	. 105	. 124
		219	. 009	. 065	. 150	. 003	. 015	. 116	. 134
		220	. 030	. 015	. 045	. 009	003	. 035	. 047
		221	. 025	. 040	. 020	. 008	. 009	. 016	. 032
		222	. 030	. 030	. 120		. 007	. 093	. 109
		223	. 020	. 060	. 115	. 006	. 014	. 089	. 109
FJ4	349								
		224	. 012	. 025	. 010	. 004	. 006	. 008	. 017
		225	. 050	. 080	. 010	. 015	. 018	. 008	. 041
		226	. 004	. 010	. 020	. 001	. 002	. 016	. 019
		227	. 004	. 025	. 025	. 001	. 006	. 019	. 026
		228	. 004	. 035	. 075	. 001	. 008	. 058	. 067
		229	. 009	. 030	. 085	. 003	. 007	. 066	. 076
		230	. 004	. 045	. 065	. 001	. 010	. 050	. 062
FJ4	420								
		231	. 011	. 020	. 005	. 003	. 005	. 004	. 012
		232	. 012	. 020	. 020	. 004	. 005	. 016	. 024
		233	. 009	. 025	. 035	. 003	. 006	. 027	. 036
		234	. 007	. 040	. 050	. 002	. 009	. 039	. 050
		235	. 007	. 015	. 065	. 002	. 003	. 050	. 056

FJALLVEDEN - Nitrogen containing ions

Table 7c

Bore- hole	Depth m	Nr	NO2 mg/1	ND3 mg/1	NH4 mg/l	NO2-N mg/1	ND3-N mg/1	NH4-N mg/1	Sum-N mg∕l
FJ8	402								
		281	<. 002	. 040	. 015	<. 001	. 009	.012	. 021
		282	<. 002	015	. 010	<. 001	. 003	. 008	. 012
		283	<. 002	. 045	. 010	<. 001	. 010	. 008	. 019
		284	. 002	. 040	. 005	<. 001	. 009	. 004	. 014
		285	< 002	. 055	. 010	< 001	. 012	. 008	. 021
		286	. 005	. 020	. 015	. 002	. 005	. 012	. 018
FJ8	562								
		287	. 004	. 045	. 015	. 001	. 010	. 012	. 023
		288	. 004	. 020	. 010	. 001	. 005	. 008	. 014
		289	. 004	. 025	. 010	. 001	. 006	. 008	. 015
		290	. 012	. 015	. 015	. 004	. 003	. 012	. 019
		291	. 004	. 020	. 010	. 001	. 005	. 008	. 014
		292	. 005	. 015	. 010	. 002	. 003	. 008	. 013
		293	. 010	. 030	. 020	. 003	. 007	. 016	. 025

FJALLVEDEN - Remaining anions and other parameters

÷											
Bore- hole	Depth m	Nr	HCD3 mg∕l	C1 mg∕1	F mg/l	SO4 mg/1	P04 mg∕1	SiO2 mg∕1	TOC: mg≠1	Turb. NTU	Drilling water residue, I%
FJ2	106										
	100	239	130	8	. 3	10	. 045	14	6		7.9%
		240	160	- 6	. 7	7.5	. 050	13	5		2.2%
		251	160	5 5	. 7	6.1		13	5		1.7%
		252	160	5	. 7.	6.1	. 045	13	5		2.0%
EJ5	293										
		253	150	8	. 6	4.0	. 14	13	6		7.9%
		254	140	8	. 6	3.2	. 16	13	6	2.5	8.7%
		255	140	8	. 6	5.2	. 18	13	చ	1.3	10.2%
		256	144	8	. 6	10	. 13	14	8	2.4	8.7%
FUR	100										
FUE	409	257	157	7	. 7	-	. 22	13	5	13	7.1%
		258	171	8	. 9	. 2 . 2	. 18	13	5	9.7	8.7%
		259	173	8	. 9	. 2	. 16	13	6	6.0	8.7%
		260	170	8	. 9	. 2	. 20	13	5	28	7.9%
		261	175	9	. 9	.2	. 14	13	4		8.7%
F J2	506										
		262	97	140	4.7	. 2	. 090	9.5	5	2.3	7.1%
		263	95	150	4.8	< 1	. 070	9.4	Ą	1.9	5.5%
		264	92	160	5.0	< 1	. 080	9.2	zi.	. 5	5 5%
		265	90	170	5.2	. 1	. 070	9.0	4	. 8 . 7	5.5%
		266	85	170	5.5	. 2	. 055	9.5	4		3.9%
		267	83	170	5.5	. 2	050	9.2	4	. 7 . 5	3.9% 5.5%
		598	81	170	5.4	. 2	055	9.4	4	. 5	0.04
		905	180	4	. 7	7.9		15			
		903	178	4	. 9	7.5		15			
		7.77	1/0	+	. 7	7.0		1.0			

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FUXLLVEDEN - Remaining anions and other parameters

Table 8b

Table 8c

Bore-	Depth	Nr	нсоз	C 1	F	504	P04	Si02	тос	Turb.	Drilling water
hole	m		mg∕l	mg / 1	mg / 1	mg∕l	mg / 1	mg∕l	mg/1	NTU	residue: I%
FJ4	131										
		211	200	8	. 6	8	. 045	7.5	7	з. з	5.5%
		212	194	10	. 6	10	. 070	8.0	7	6.9	4.9%
		213	185	10	. 6	14	. 050	8.5	7	6.3	4.3%
		214	178	10	. 6	10	. 010	10	7	6.3	4.4%
		215	218	6	1.0	7	. 025	5.8	6	3.3	6.0%
		216	207	7	. 9	8	. 010	6.7	5	2.4	5.2%
FJ4	272										
101	ton / ton	217	192	9	. 6	8	. 085	8.3	a	6.8	4.1%
		218	190	9		7	. 075	8.3	8	9.4	4, 4%
		219	193	9	. 6	7	. 075	8.4	7	6.0	4. 4%
		220	191	9	. 6	5	. 130	8.5	н н	11	4.1%
		221	196	9	.5	5	. 055	8.1	7	6.9	4.7%
		222	196	9	. 5	7	. 045	8.2	7	2.2	4.4%
		223	196	9	. 6	7	. 035	8.4	7	5.1	4.4%
FJ4	349										
FQ 1	317	224	205	13	1.9	<. i	. 29	7.8	40	4.5	7.6%
		225	200	13	2.2	< 1	. 31	7.7	30	4.6	7.9%
		226	195	13	2.5	< 1	. 26	7.5	20	5.2	7.7%
		227	194	11	2.7	<. 1	. 24	7.5	15	5.7	7.9%
		228	194	5	1.1	2.6	. 115	6.7	6	1.5	3.4%
		229	195	5	1.0	3. 6	. 105	6.6		2.4	3.3%
		230	197	1	1.0	Э. 4	. 070	6.7	7	2.2	4.3%
FJ4	420										
	120	231	110	220	3.8	. 2	. 030	5.5	50		4.8%
		232	162	30	2.5	3.7	. 050	5.4	25		2.2%
		233	185	14	2.0	3.8	. 070	6. i	15		1.3%
		234	198	8	1.7	3.9	. 060	6.2	8		. 7%
		235	203	6	1.6	3.5	060	6.2	7		. 6%
			200	5		2. 0					
		904	184	4	. 9	7.3		14			
		904	184	4	. 7	7.9		15			
		700	104	-	. /	1.7		10			

FJALLVEDEN - Remaining anions and other parameters

Bore-	Depth	Nr	нсоз	C 1	F	S04	P04	Si02	TOC	Turb.	Drilling water
hole	m		mg∕l	mg / 1	mg⁄1	mg∕l	mg / 1	mg∕l	mg / 1	NTU	residue, I%
FJB	402										
		281	123	4	1.0	7.5	. 015	7.8	3	4.4	5. 5%
		282	123	4	1.0	7	. 015	7.9	4	26	5. 5%
		283	128	4	1.0	6.5	. 020	8.0	3	20	6.3%
		284	128	4	1.0	6	. 020	8.3		22	
		285	130	4	1.0	6.5	. 025	8.4	3	22	5.5%
		286	130	4	1.0	6	. 020	8.4	3	6.7	5.5%
FJB	562										
		287	128	4	. 9	6	. 025	8.2	3	11	5.5%
		288	130	4	. 9	6	. 030	8.2	З	6.0	5. 5%
		289	130	4	. 9	6	. 040	8.2	3	11.5	5. 5%
		290	130	4	. 9	6	. 045	8.7	3	9.0	4.7%
		291	132	4	. 9	6	. 050	8.4	3	11	3.9%
		292	130	4	. 9	5	. 050	8.5	З	11	4.7%
		293	129	4	. 9	5	. 055	8.5	з	8.5	7.1%
		908	140	3	. 4	6.9		18			
		909	137	3	. 4	7.0		17			

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Bore-	Depth	Nr	Sample	S, prt	Fe,prt	Al,prt
hole	m		ml	mg/l	mg∕l	mg∕l
FJ2	106	239	1900	. 002	. 074	. 095
FJ2	106	252	1700	. 003	. 032	. 112
FJ2	293	255	500	. 006	. 128	. 300
FJ2	293	256	500	. 012	. 118	. 220
FJ2	409	257	350	. 026	. 371	. 237
FJ2	409	260	1650	. 002		. 039
FJ2	506	265	2000	. 003	. 019	. 032
FJ2	506	267	950	. 004	. 103	. 021
FJ4 FJ4	131 131	213 215	600 850	012	. 133 . 032	. 267
FJ4	272	555	350	. 023	1.829	. 157
FJ4	272	550	425	. 009		. 329
FJ4 FJ4	349 349	227 229	300 2250	047	270	. 667 . 017
FJ4 FJ4	420 420	231 234	1200 1950	. 018	. 242	. 258 . 042
FJB	402	282	2230	. 012	. 356	008
FJ8 FJ8	562 562	290 292	2250	. 004 . 004	016	004

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Table 10

FJALLVEDEN - Uranium, thorium, radium and radon

Bore-	Depth	Nr	Th	U	Ra-226	Rn-222
hole	m		ug/l	Bq/l	Bq/1	Bq/1
FJ2	106	252	(2.6+0.6)E-2	(1.4+0.4)E-2	(5.5+0.8)E-3	201+5
FJ2	293	256	(9.2+4.5)E-2	(6.1+3.2)E-3	(5.5+1.0)E-3	266+5
FJ2	409	260	(1.2+0.4)E-1	< 2.3 E-3	< 5.8 E-3	120+5
FJ2	506	267	< 2.6 E-1	< 5.1 E-3	(3.5+0.3)E-3	305+6
FJ4	131	215	<pre>< 1.2 E-2 (4.4+1.8)E-3 (1.9+0.6)E-2 (2.5+0.6)E-2</pre>	(9.3+0.5)E-2	< 3.9 E-3	151+5
FJ4	272	222		(7.7+0.5)E-2	< 3.0 E-3	240+5
FJ4	349	229		(1.6+0.3)E-2	(8.5+1.3)E-3	166+5
FJ4	420	234		< 4.8 E-3	< 1.1 E-3	189+5
FJ8	402	285	< 3.8 E-2	(2.8+0.4)E-2	(2. 9+0. 6)E-3	144+5
FJ8	562	292	< 6.9 E-2	(1.1+0.3)E-2	(3. 1+0. 8)E-3	107+5

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- TR 83-01 Radionuclide transport in a single fissure A laboratory study Trygve E Eriksen Department of Nuclear Chemistry The Royal Institute of Technology Stockholm, Sweden 1983-01-19
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- TR 83-04 Stability of bentonite gels in crystalline rock -Physical aspects Roland Pusch Division Soil Mechanics, University of Luleå Luleå, Sweden, 1983-02-20
- TR 83-05 Studies in pitting corrosion on archeological bronzes - Copper Åke Bresle Jozef Saers Birgit Arrhenius Archaeological Research Laboratory University of Stockholm Stockholm, Sweden 1983-01-02

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- TR 83-12 Calculation of activity content and related properties in PWR and BWR fuel using ORIGEN 2 Ove Edlund Studsvik Energiteknik AB Nyköping, Sweden 1983-03-07
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- TR 83-14 The complexation of Eu(III) by fulvic acid J A Marinsky State University of New York at Buffalo, Buffalo,NY 1983-03-31
- TR 83-15 Diffusion measurements in crystalline rocks Kristina Skagius Ivars Neretnieks Royal Institute of Technology Stockholm, Sweden 1983-03-11
- TR 83-16 Stability of deep-sited smectite minerals in crystalline rock - chemical aspects Roland Pusch Division of Soil Mechanics, University of Luleå 1983-03-30
- TR 83-17 Analysis of groundwater from deep boreholes in Gideå Sif Laurent Swedish Environmental Research Institute Stockholm, Sweden 1983-03-09

TR 83-18 Migration experiments in Studsvik O Landström Studsvik Energiteknik AB C-E Klockars O Persson E-L Tullborg S Å Larson Swedish Geological K Andersson B Allard B Torstenfelt Chalmers University of Technology 1983-01-31

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- TR 83-19 Analysis of groundwater from deep boreholes in Fjällveden Sif Laurent Swedish Environmental Research Institute Stockholm, Sweden 1983-03-29
- TR 83-20 Encapsulation and handling of spent nuclear fuel for final disposal l Welded copper canisters 2 Pressed copper canisters (HIPOW) 3 BWR Channels in Concrete B Lönnerberg, ASEA-ATOM H Larker, ASEA L Ageskog, VBB May 1983
- TR 83-21 An analysis of the conditions of gas migration from a low-level radioactive waste repository C Braester Israel Institute of Technology, Haifa, Israel R Thunvik Royal Institute of Technology November 1982
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- TR 83-24 Corrosion resistance of a copper canister for spent nuclear fuel The Swedish Corrosion Research Institute and its reference group Stockholm, Sweden April 1983
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- TR 83-27 Radiation effects on the chemical environment in a radioactive waste repository Trygve Eriksen Royal Institute of Technology, Stockholm Arvid Jacobsson University of Luleå, Luleå Sweden 1983-07-01

ANALYSIS OF GROUNDWATER FROM DEEP BOREHOLES IN FJÄLLVEDEN

Sif Laurent

IVL, Swedish Environmental Research Institute Stockholm, Sweden 1983-03-29

This report concerns a study which was conducted for SKBF/KBS. The conclusions and viewpoints presented in the report are those of the author(s) and do not necessarily coincide with those of the client.

A list of other reports published in this series during 1983 is attached at the end of this report. Information on KBS technical reports from 1977-1978 (TR 121), 1979 (TR 79-28), 1980 (TR 80-26), 1981 (TR 81-17) and 1982 (TR 82-28) is available through SKBF/KBS.

ANALYSIS OF GROUNDWATER FROM DEEP BOREHOLES IN FJÄLLVEDEN

Sif Laurent

IVL, Swedish Environmental Research Institute Stockholm, 1983-03-29

SUMMARY

Groundwater from three boreholes at an investigation site in Fjällveden has been sampled and analyzed. This is part of a larger programme of geological, geophysical and hydrogeological investigations aimed at finding a suitable site for a high level radioactive waste repository.

Four water-bearing levels in each of the boreholes FJ2 and FJ4, and two levels in borehole FJ8, have been sampled. The sampling depths range from 106 down to 562 m. Prior to sampling, the water-bearing section is isolated between packer sleeves. The water is then pumped to the surface where sensitive parameters such as redox potential, pH, sulphide and oxygen content are measured electrochemically on the flowing water in a system isolated from the air. Water, filter and gas samples are sent to several laboratories for further analysis.

The present report is a presentation of the results of the groundwater analyses. The reliability of the results is discussed but there is no evaluation in relation to geology and hydrogeology. This report presents the basic results from the groundwater analyses to be further evaluated by experts in different fields.

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INTRODUCTION

Groundwater from three boreholes in Fjällveden has been sampled and analyzed. The sampling levels were chosen on the basis of hydrological measurements in the boreholes. Sampling was carried out by Sveriges Geologiska AB. Personnel from IPK and VIAK performed the field measurements and sampling. Several laboratories participated in the water analysis (see below). Supplementary sampling in additional boreholes in Fjällveden is in progress. The results will be reported at a later time.

The boreholes were pumped out with a gas lift prior to the hydrological investigation and immediately before the equipment for water sampling was lowered to the first sampling level. On the latter occasion, three such gas lift pumpings - called mammoth pumpings were performed in a row. Iodide (NaI, 0.01 mmol/1) was added to the drilling water during drilling to enable any residual drilling water to be traced in the groundwater. The selected fracture zones were sealed off by packers spaced at a distance of 2.7 m.

Sampling took place from the middle of June to the middle of October, 1982. Four levels were sampled in each of the boreholes FJ2 and FJ4, and two levels in FJ8. Depth is given both as vertical depth (depth) and/or borehole length (length). It is always vertical depth that is given, unless otherwise specified in the table heading.

The work was carried out in periods of 14 days - lowering and start of the sampling pump, 4 days off and 9 days field measurements. When the supply of water is adequate, approximately 4.5 cubic metres of water can be pumped up during one period.

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The present report consists primarily of a presentation of the results of the groundwater analyses. The reliability of the results is subjected to some scrutiny. There is, however, no evaluation in relation to geology or hydrology. The material in this report will be further analyzed by experts in different fields.

The geology and hydrology of the test area will be described in KBS TR 83-52.

A general description of the chemistry of groundwater from great depths in granite and gneiss has previously been published by G Jacks (KBS TR 88). The same author has also described the chemistry of groundwater in Blekinge (KBS TR 79-07), where Sternö is situated. The results of analyses of the groundwater from the Finnsjö area are presented in KBS TR 82-23 and from Gideå in KBS TR 83-17. A geochemical evaluation of fracture minerals and the relation between fracture minerals and groundwater composition in Finnsjön has been published by E-L Tullborg and co-workers (KBS TR 82-20). The relationship between pH and carbonate content in deep groundwaters has been discussed by Bert Allard in KBS TR 82-25. The expected groundwater composition and its importance for the final storage of radioactive waste have been discussed in KBS TR 90 and the final report KBS-2 Volume 2 (Handling and Final Storage of Unreprocessed Spent Nuclear Fuel, Technical Volume). Hydrology and groundwater age are also taken up in the latter report.

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SAMPLING

2.1 Drilling water

Some of the drilling water that is used in the core drilling of the sampling holes can penetrate into rock fractures and contaminate the groundwater there. In order to get rid of the drilling water to as great an extent as possible prior to sampling, three mammoth pumpings (using nitrogen gas) are performed immediately prior to installation of the sampling equipment. The sampling equipment (2.2) is lowered to the chosen level and allowed to work for 3-4 days before sampling begins. Periodic sampling during a period of 8-9 days makes it possible to follow changes in the composition of the water that might derive from drilling water or some other source of contamination. The changes may also be due to the incursion of water from different aquifers.

Water from percussion-drilled boreholes has been used as drilling water. The drilling water was filtered through mechanical filters of cellulose (18 CMC 3-2), which are supposed to retain particles with diameters larger than $5 \,\mu$ m. The filtered drilling water was then analyzed. The results for the boreholes in question are presented at the bottom of the respective tables (tables 5-8).

Before the drilling water was used, it was "marked" by the addition of sodium iodide (0.01 mmol/l in the drilling water) in order to make sure that it would be possible to trace any drilling water that may have contaminated the sampled water. Unfortunately, it has been found that both the drilling water and sampled groundwater can have considerable contents of natural iodide (KBS TR 83-17). Chloride, for example, is always accompanied by small quantities of bromide and iodide. If we assume 3×10^{-4} mg I per mg Cl, which is a value that has been observed in connection with water sampling at Stripa, chloride concentrations of 100-200 mg/l correspond to iodide concentrations of 0.03-0.06 mg/l, which in turn would be erroneously interpreted as a drilling water contamination of 2-5%. The drilling water contamination given for FJ2, 506 m should for this reason be 2-4% too high, and the drilling water contamination in the first sample (No. 231) in FJ4, 420 m can probably be explained by the high salt content. The drilling waters used in Fjällveden were not analyzed for iodide prior to marking, which brings an additional uncertainty to the drilling water contaminations given in table 8.

2.2 Sampling equipment

The boreholes are core-drilled with a diameter of 56 mm. The 2.7 m long sampling zone is sealed off by rubber packers that are expanded

to a pressure that is 0.8-1 MPa above the groundwater pressure. The sampling pump is positioned immediately above the sealed-off zone. The intake is from the upper part of the zone. The pump, which is made of steel and furnished with teflon seals, has a maximum capacity of 0.12 l/stroke, equivalent to 330 l/d. The water flow available for sampling decreases if the flow of water in the measured rock is lower than the capacity of the pump. A schematic illustration of the sampling pump and packers is shown in Fig. 1.

The water is pumped up through teflon-lined iron pipes to a test chamber of stainless steel on the ground surface. The test chamber is equipped with electrodes and measuring cells for recording pH, Eh (glassy carbon), Eh (platinum), pS and the oxygen content and conductivity of the water. The test chamber is also equipped with valves for the extraction of water samples.

As of September, the equipment has been augmented with an extra measuring chamber of glass with two Eh electrodes of the same kind as those in the test chamber (glassy carbon and platinum). In order to give these electrodes plenty of time to stabilize, they are not calibrated and they are only cleaned when the level is changed.

2.3 Procedure

Work at the boreholes is conducted in periods of 14 days. On the last day of a period, the sampling equipment is set up at the level that is to be sampled during the following period, and the sampling pump is started. The personnel then go home for 4 days off while the sampling pump works. Sampling then proceeds for 10 consecutive days. There have been some disruptions in the programme, as is evident from the notes to the table in 2.4.

2.4 Water flow during sampling

The water flow at different levels is shown in the table below. The pumped-up water volume has also been calculated. "Prior" refers to

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the approximate quantity of water pumped up from the level before sampling started, and "Sampl." to the quantity of water during the sampling period.

BH	Depth m	Mean flow 1/day	Prior m ³	Samp1.	Total m ³	Notes
FJ2	106	270	0.0	1 5	1 7	-1
£JZ	293	245	0.2	1.5	1.7	1
				2.2	_	2
	409	265	-	1.6	-	3
	506	190	1.1	1.7	2.8	
FJ4	131	315	1.9	2.5	4.4	
	272	320	1.9	2.6	4.5	
	349	335	2.0	2.7	4.7	
	420	260	0.2	1.6	1.8	4
FJ8	402	250	0.0	1.8	1.8	5
	562	145	0.9	1.3	2.2	

Note 1 Due to delays at the borehole, sampling started the day after lowering.

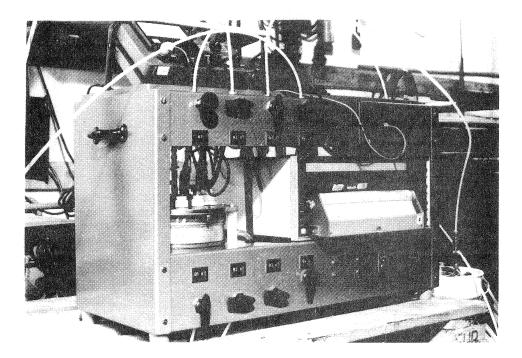
- Note 2 The pump ran for about seven days prior to sampling, but unfortunately there was no pressure in the packers when the personnel returned after their leave. It is therefore probable that water leaked into the sampling zone.
- Note 3 Due to difficulties in connection with lowering, the pump was not started until 14 h before the start of the measuring period. The pressure in the packers was lost again in that time, so water may have leaked into the sampling zone.
- Note 4 Here again, difficulties were encountered in connection with lowering. The sampling zone, which is only sealed off by a single packer, goes from the bottom of the borehole at 585 m to 420 m. The sampling pump started the day before the measurements.

Note 5 The pump was started 2.5 hours before the readings began. The 1st sampling was done 24 hours later.

3 FIELD MEASUREMENTS

The field measurements are recorded in tables 1-3, the E^{O} values obtained from the field calibrations in table 4 and the field values obtained during water sampling in table 5.

The photograph below shows the field equipment with test chamber, measuring equipment and valves for water sampling.



3.1 Calibrations

Each calibration has been assigned a calibration number. Measurements made after the calibration have the same number.

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All calibration solutions are freshly prepared from concentrate and deaerated distilled water, except the one used for calibration of the conductivity cell.

Three buffer solutions with pH 4, 7 and 10 are used for calibration of the pH electrode. Quinhydrone is added to the buffer solutions with pH 4 and 7 for simultaneous calibration of the Eh electrodes. Concentrate diluted to solutions that are 0.01 and 0.05 molar with respect to sulphide ion is used for calibration of the sulphide electrode. The E^{0} values obtained are presented in table 4.

The oxygen probe is calibrated against air-saturated distilled water and the conductivity cell against 0.01 and 0.1 mol/l KCl.

3.2 Measurement results

Each measuring occasion is identified with a calibration number, borehole designation, length of borehole (core length) as well as day and time. The vertical depths corresponding to the borehole lengths are shown in the table below.

Bore-	Length	Depth	Bore-	Length	Depth	Bore-	Length	Depth
hole	m	m	hole	m	m	hole	m	m
FJ2	123 342 483 605	106 293 409 506	FJ4	151 317 410 496	131 272 349 420	FJ8	470 666	402 562

Owing to the fact that it takes some time after each calibration before the water in the test chamber is representative of the borehole water and before the Eh and pS values have stabilized, no measurement results have been included until at least 24 hours after the calibration - in general, from and including the second day after calibration. Nor are values included that have been obtained from obviously unsuccessful measurements owing to problems encountered in the field. Due to a fault in the measuring equipment at FJ4, pH values were only obtained from the measurements during the first days on the upper level.

Stable measurement results were not obtained from the extra Eh electrodes that were connected before the test chamber until voltmeters with higher internal resistance were procured. The values measured with the extra electrodes are marked "ex" in table 3.

3.3 Temperature measurement

The temperature is measured in the test chamber to permit calibration and calculation of the electrode potentials.

Owing to the relatively slow rate of water turnover in the test chamber, the temperature measured is not representative of the groundwater. It is therefore not reported.

3.4 Measurement of pH, Eh and pS

The system often seems to need to remain undisturbed for about a week before the Eh and pS values are more or less stable. The pH value stabilizes much faster.

The calibrations show that adjustments must be made fairly frequently, probably because a deposit builds up on electrodes and probes. The optimum time between two calibrations would therefore appear to be between 5 and 10 days. The electrodes were usually cleaned when they were lowered to a new level as well as before calibrations 203, 204, 207, 208, 304, 305, 306, 403 and 406.

Due to an amplifier malfunction, almost all pH values for FJ4 are lacking.

3.5 Oxygen measurement with probe

When the oxygen probe was recalibrated, it was found that the measuring error was 5-25% - in the negative direction, as a rule. In seven cases, errors greater than 40% were found, and the probe had to be cleaned and the membrane changed. Values before such a major adjustment are not reported.

It appears as if the oxygen probe has to be recalibrated more frequently than is possible according to 3.4 in order for reliable results to be obtained.

Despite the difficulties encountered in oxygen measurement, the field values often show good correlation with pS; high oxygen content - high pS (low sulphide content).

3.6 Measurement of conductivity

Calibration was only done at the beginning of each level. The conductivity measurements were consistently stable.

4 MAIN COMPONENTS OF THE WATER

Each sampling occasion has been assigned a unique sampling number, which is the same for all samples taken on the same occasion ("Nr." in the tables).

The main components were analyzed by SGAB's water laboratory in Uppsala and by Hydroconsult in Stockholm. Some control analyses were performed by IVL (the Swedish Institute for Water and Air Polution Research) in Stockholm and IMLAB in Linköping. The analysis results are reported in tables 5-8. The table below shows which laboratory carried out the analyses and which method was used. Where there are two analyses of the same parameter, the mean value is reported.

	SGAB	H-con.		SGAB	H-con.
		· · · ·			
$\mathbf{p}\mathbf{H}$	*	*	HCO ₃	g	Ъ
cond.	*	*	C1 ³	i	h
Turb.		*	F	i	k
Ca	а	Ъ	SO,	i	
Mg	а		PO,4		f
Na	а	с	NO_{3}^{4}		f
K	а	С	NO		f
Mn	а	d	NH ²		f
Fe(II)	е		sið,	m	m
Fe-tot	а	,	s ^z	n	
A1	а		I	0	
Cu	а		TOC	р	q
Sr	а			-	-

* Conventional according to Swedish Standard (SS)

a Optical emission with ICP (own design)

b Titrimetric with EDTA according to SS

c Emission with flame

d Atomic absorption

e Colorimetric with o-phenanthroline according to Standard Methods

f Colorimetric according to SS

g Specially designed titrator

h Titration according to Mohr (SS), high concentrations only

i Ion chromatograph, for chloride, low concentrations only

k Potentiometrically according to SS

- m Colorimetrically with methylene blue according to Standard Methods
- n Potentiometrically
- Colorimetrically, higher concentrations also potentiometrically
- p Carbon analyzer from Carlo-Erba
- q Carbon analyzer from Oceanographic

Drilling water was analyzed by SGAB. The analyses are presented at the bottom of tables 5, 6 and 8.

The field values obtained simultaneously with the water sampling and the laboratory analyses of the same parameters are presented in table 5. Note that the field values, which are measured during water sampling, are often not representative of the borehole. Sampling may, for example, have been done immediately after a calibration.

The date is given for both the sampling day and the day the samples were received at the laboratories. In both cases, the date is given with the number of the week followed by the number of the day of the week. (Day 1 is Monday.)

All determinations of metal ions are presented in table 6, while table 8 shows TOC, turbidity and the negative ions that do not contain nitrogen.

The parameters that contain nitrogen are compiled in table 7. The analysis results are given as the concentration of the ion and the concentration of nitrogen.

"Drilling water residue" (table 8) is the percentage fraction of the original iodide concentration in the drilling water (2.1) that has turned up in the groundwater. No correction is made for the water's natural iodide content (2.1).

4.1 Sampling

The water samples were taken through a valve in immediate connection with the test chamber. On each sampling occasion, a new tube, kept well protected from dust, was attached to the valve. The free and untouched end of the tube was inserted down to the bottom of the sample bottle. At least two bottle volumes of sample water were allowed to run over before the bottle was sealed, without any air having been trapped inside. The samples were kept in a refrigerator awaiting transport (4.2). 4.1.1 Unpreserved samples for analysis of negative ions etc. were taken in bottles of borosilicate glass (1 litre) with a ground, filled stopper held in place with a steel clip.

In order to keep the ground surface moist so that no air can leak through, the bottles were stored and transported upside-down.

4.1.2 Preserved samples for analysis of metal ions were taken in acidwashed polyethylene bottles (250 ml).

The samples were preserved with 2.5 ml of concentrated hydrochloric acid (super-pure). The acid was added through a dispenser below the sample surface immediately before the sample bottle was sealed. Hydrochloric acid was chosen as a preservative because iron(II) was to be determined.

4.2 Transport of samples

The samples were packed in specially-made insulated boxes with frozen blocks. The boxes were mailed "express" on the afternoon of the sampling day and delivered to the laboratory the following morning. No water sampling took place on Fridays and Saturdays, since the samples could then not be received by the laboratory until Monday or Tuesday. The Sunday samples were sent together with the Monday samples.

4.3 Sampling levels

Field analyses representative of the borehole are presented in tables 1-3, while the field analyses in table 5 are only representative of the sampling occasion.

Except for the redox-sensitive parameters, the analysis results are, as a rule, strikingly consistent within the levels in FJ2 and FJ8. In FJ2, 106m, as well as in all of FJ4, pronounced trends are found in the concentrations of anions and cations, least pronounced in FJ4, 272m where only iron, phosphate and nitrogen-containing ions change. There is a possibility that water has leaked into the sampling zone due to a pressure loss in the packers immediately before the start of sampling at levels FJ2, 293m and 409m. No appreciable effect of the leakage is noticeable. The analyses within the levels are uniform and without any trend.

The highest salt concentrations are found in FJ2, 506m and in the first sampled water from FJ4, 420m.

4.4 pH

pH was determined directly in the field and by both laboratories (table 10). The field values are systematically around 5% h gher than the laboratory values. A similar difference has been noted in many groundwaters. There is also a small systematic difference between the laboratories, which may be attributable to the fact that Hydroconsult, which notes the lower value, thermostated its samples at 25° C before measurement.

The differences within the levels are small, as a rule. The highest pH values were measured at the deep levels.

4.5 Conductivity

Conductivity was also measured both in the field and at two laboratories. As a rule, the field value is slightly higher than the laboratory values, but the differences are relatively small. An exception from this rule is FJ4, 420m, where the field value is only about half the laboratory values, which agree closely with each other. The conductivity meter in the field was probably set wrong.

4.6 Organic carbon (TOC)

No parameter exhibits such large differences between the laboratories as TOC - SGAB's results are often more than twice Hydroconsult's results. Control analysis at two Swedish laboratories comes closer to verifying SGAB's high results, while analysis at a laboratory in the United States, which specializes in the determination of humic and fulvic acids, confirmed Hydroconsult's low values. The mean value is reported to only one decimal place (table 8).

The normal concentration lies between 4 and 8 mg/l in FJ2 and FJ4, while it is only about 3 mg/l in FJ8. The levels FJ4, 349m and 420m have very high concentrations at the start of the pumping period, after which the concentrations gradually decline to normal values. Other ions exhibit similar trends at these two levels.

4.7 Negative ions

The negative ions are compiled in tables 7 and 8. The results generally show good consistency within the sampling zones. Where trends occur (4.3), the change is verified by other ions.

Very high nitrogen concentrations, mainly as nitrate, are found in the two middle samples in FJ4, 131m. They have no counterpart in any equally pronounced change in any of the cations analyzed. Agreement between the laboratories' N analyses is good for these two samples. pH and conductivity are determined in the same sample bottle. Normal values are reported by both laboratories.

4.8 Positive ions

The positive ions are compiled in table 6. The concentrations generally differ very little between water samples from the same level. Where trends do exist (4.3), they are verified by changes in several ion concentrations.

4.8.1 Lanthanum, neodymium and ytterbium were determined in the same samples as strontium. In all cases, the concentrations were below the detection limit (< 0.005 mg/l).</p>

5 PARTICULATE MATTER

The particulate matter was subjected to several different types of analyses (5.1-5.5).

5.1 Chemical composition

The water was filtered through a membrane filter (Millipore, pore diameter 0.45 μ m) installed directly after the test chamber. The filtered volume was between 300 and 2300 ml. The total concentration on the membrane was determined by X-ray fluorescence at SGU's (the Geological Survey of Sweden) laboratory in Luleå. The concentrations are presented in table 9 as mg/l of filtered sample. The s. ple volume is also given in the table. The quantity of copper, strontium, lanthanum, neodymium and ytterbium was, with the exception of samples Nos. 260 and 282 (where 1 and 2 μ g Sr, respectively, was obtained), below the detection limit (1, 1, 3, 2 and 2 μ g respectively).

There is a clear correlation between high volume and low concentration, which is natural since the amount of filtrate that can pass a filter is normally dependent on the amount of particles that stay on the filter. The iron and sulphur concentrations also covary, which is not unexpected either.

5.2 Turbidity

Turbidity (table 8) was determined for all samples except those from FJ2, 106m and FJ4, 420m.

5.3 Particle distribution

The particle distribution within the range 2-80 μ m was determined at all levels. The determination was performed by VIAK's water pollution research laboratory in Stockholm. NOTE! It is the length of the borehole and not its vertical depth that is indicated on the curves.

5.4 Humic and fulvic acids

The two deepest levels in the boreholes were analyzed. The determination was performed at Batelle, USA, through combined dialysis and gel film chromotography (GFC). The method, which fractionates the sample into humic acids and four molecular weight fractions of fulvic acid, is described by Means et al 1977 (Limnol. Oceanogr., 22, 957-965). Sodium tetraborate (pH 9.1) was used on recommendation by Swift and Posner 1971 (J. Soil Science, 22, 237-249).

Owing to the samples' low content of organic matter, they were first concentrated ten times before analysis. No loss of volatile organic matter was found in connection with the concentration process.

Level	No.	>1000	700-100	300-700	<300
FJ2, 409m	260	28	63	8	1
FJ2, 506m	266	18	63	15	4
FJ4, 349m	229	48	47	5	0
FJ4, 420m	234	53	37	9	1
FJ8, 402m	285	12	67	18	3
FJ8, 562m	292	23	46	1.7	14

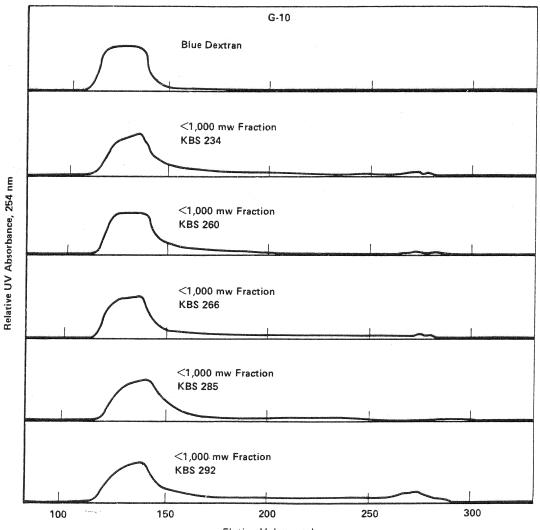
No humic acids were present in the samples. The samples' contents of different molecular weight fractions of fulvic-acid-like substances are shown in the table below. The eluates' UV absorbance, recorded continuously at 254 nm, is shown on the next page. According to Means, the results strongly indicate that the most important organic constituents are hydrophilic polymer substances with low to medium molecular weights - probably fulvic acids or similar substances, which are the predominant substances in natural water.

ISOTOPE ANALYSIS

6

Isotope analysis of light substances in the water is done mainly to permit calculations of the age and origin of the water. The heavy radioactive substances are determined primarily because the natural background concentrations are of great interest for the KBS project.

The analyses were carried out at the following laboratories: Laboratory for Isotope Geology, Stockholm (6.1, 6.3) Institute for Radiohydrometry, Neuherberg, West Germany (6.2, 6.4) Mass Spectrometry Laboratory, Uppsala University (6.4) Studsvik Energiteknik, Nyköping, Sweden (6.5)



6.1 Carbon isotopes

Prior to the determination that is used for dating by means of the carbon-14 method, the water's contents of carbon dioxide and carbonate have to be concentrated. It is thereby of the utmost importance that the concentrate not be contaminated by chemicals or contact with air.

6.1.1 Sample preparation is done in the field. A polyethylene barrel holding 130 l and filled from the beginning with nitrogen is filled with water. Hydrochloric acid is added to disintegrate carbonates present in the water. A nitrogen gas stream is then used to drive the carbon dioxide over to a wash bottle containing sodium hydroxide.

6.1.2 Groundwater age is presented in the table below as "Age BP" and age after correction for C-13 content as "Age BP, corr".

Bore- hole	Depth m	Nr	Age BP	Age BP corr	C13 0/00
FJ2	106	252	4635	4725	-19.5
FJ2	293	256	11010	10960	-28.0
FJ2	409	260	4145	4235	-19.3
FJ2	506	267	13665	13920	- 9.5
FJ4	131	215	3795	3950	-15.3
FJ4	272	222	3345	3475	-17.0
FJ4	349	229	5375	5535	-15.0
FJ4	420	234	6660	6850	-13.3
FJB	402	285	3880	3980	-18.7
FJB	562	292	3890	3975	-19.9

6.2 Deuterium and oxygen

The analyses can furnish information on the climatic conditions prevailing at the time of infiltration.

Bore- hole	Depth m	Nr	H-2 (D)	0-1(3 (D)	9-18 (S)
FJ2	106	252	-80.5	-11.33	11 24
FJ2	293	256			-11 31
FJ2	409	260			-11, 35
FJ2	506	267			-14.11
FJ4	131	216			-11.50
FJ4	272	222	-82.6	-11.54	11.57
FJ4	349	229	81.6	-11.45	-11.73
FJ4	420	234	-84.7	-11.69	t t . 88
FJ8	402	285	-79.3	-11.22	11, 21
FJB	562	292	-77.8	-10.94	-11.16

The concentrations in the table refer to deviations in per mill from SMOW (Standard Mean Oceanic Water). The analyses were performed in West Germany (D) and in Sweden (S).

6.3 Tritium

Owing to its short half-life (about 12 years), tritium is an important isotope in hydrologic studies. The amount of tritium in the atmosphere has increased drastically due to nuclear weapons testing. The tritium content of rain water has increased by more than ten times, which makes it possible to determine whether "young" water is present in a groundwater sample.

The tritium concentration is given in the tables in the unit TU, which is the number of tritium atoms per 10^{18} hydrogen atoms.

No tritium determination has been carried out on the drilling water used here.

19

Nr	Bore-	Depth	Tr
and the second	hole	m	TU
1			
256	FJ2	293	19
260	FJ2	409	19
267	FJ2	506	<3
215	FJ4	131	9
222	FJ4	272	21
229	FJ4	349	12
234	FJ4	420	6
285	FJ8	402	8
	FJB	562	10
Ross / Loss	8 3. See 9	Inf and Seno	

6.4 Uranium, thorium, radium and radon

The natural concentration of these elements in the groundwater is of great interest for the project. The analyses are presented in table 10. The concentrations for U, Ra and Rn are given in Bq/1. The following relationships apply:

39.4 μ g U per Bq 2.74 x 10⁻⁵ μ g Ra-226 per Bq 1.76 x 10⁻¹⁰ μ g Rn-222 per Bq

GAS ANALYSIS

7

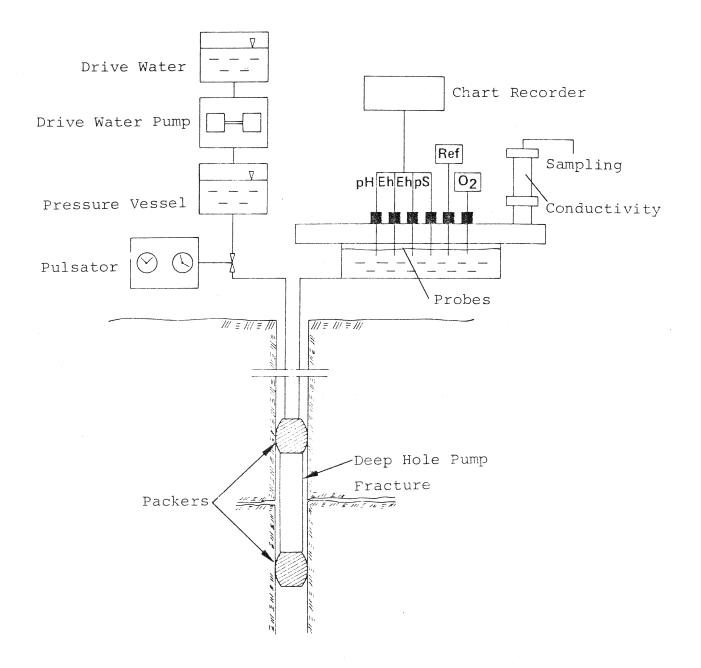
Helium was determined in the gas that spontaneously leaves the water on its passage through a Horst bottle. The analysis was carried out by AGA SpecialGas, Lidingö, Sweden. No gas sample was taken from FJ4, 349m.

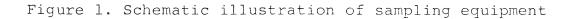
In the table below, besides the percentage content of helium in the gas phase, the amount of water that has passed through the Horst bottle and the obtained gas volume are also given. This enables the reader to form his own rough idea of the quantity of dissolved gases in the water. The helium content of the water has not been calculated, since degassing is dependent on a number of uncontrolled parameters. We know, for example, that temperature variations of >20 °C occur in the tent where the borehole opens out and the sampling equipment is located.

BH	Depth m	No.	Helium %	Water volume 1	Gas volume m ³	Notes
FJ2	106	252	0.099	218	90	
	293	256	0.002	217	120	
	409	260	0.008	103	145	
	506	267	1.1	19	190	
FJ4	131	215		122	185	2
	272	222	0.004	240		1
	420	234	0.005	195		1
FJ8	402	285	0.003	185	90	
	562	292	0.004	84		1

Note 1 Data on gas volume lacking.

Note 2 Gas analysis lacking.





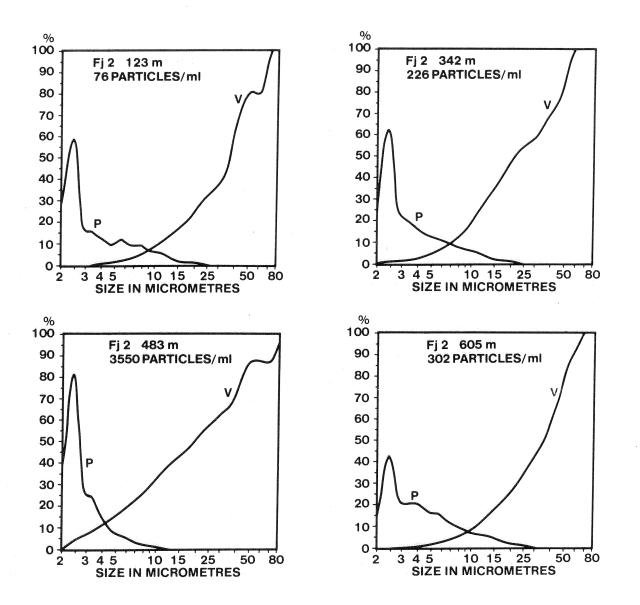


Figure 2a. Particle size distribution analysis in the region $2-80~\mu$ m. An HIAC PA 500 instrument is used. P represents the percent of the total number of particles with a certain particle size. V represents the percent of the total numer particle volume less than a certain particle size.

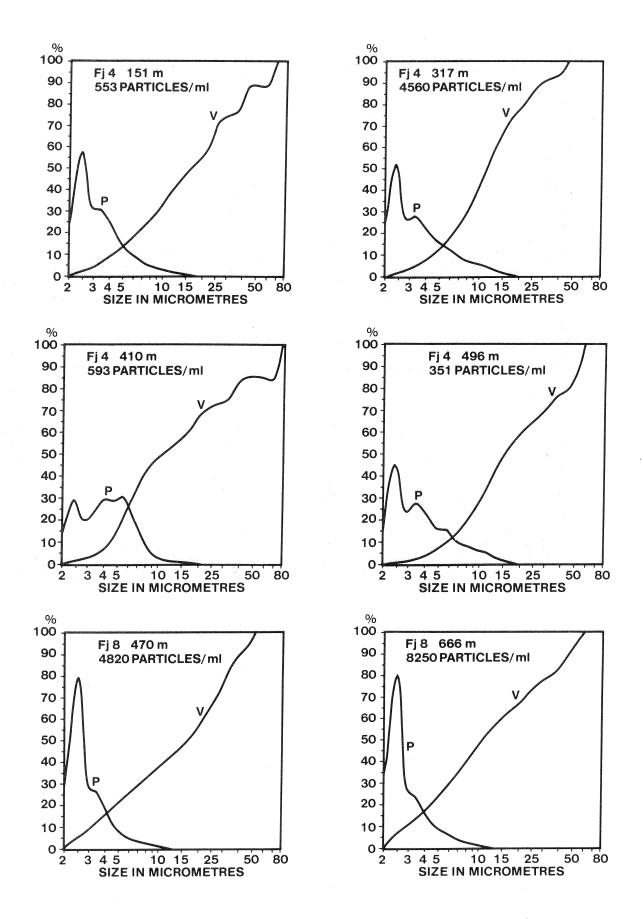


Figure 2b. Particle size distribution analysis in the region 2-80 μ m. An HIAC PA 500 instrument is used. P represents the percent of the total number of particles with a certain particle size. V represents the percent of the total particle volume less than a certain particle size.

FJALLVEDEN - Field measurements

Table 1

Cali- bration nr	Bore- hole	Hole- length m	Date ww-d	Time	Conduc- tivity mS/m	Oxy− gen mg/l	pН	Eh,C mV	Eh,Pt mV	βĘ
301	FJ2	123	34-6 34-6 34-6 34-6 34-7 34-7 34-7 34-7 34-7 34-7	830 1030 1230 1330 1530 745 930 1130 1230 1330	26. 9 26. 9 26. 9 26. 9 26. 9 27. 0 27. 0 27. 0 27. 0 27. 0 27. 0	. 00 . 00 . 00 . 00 . 00 . 00 . 00 . 00	8.0 8.0 8.0 8.0 8.0 8.2 8.1 8.1 8.2 8.1	- 59 - 62 - 65 - 66 - 88 - 88 - 88 - 88 - 88 - 88 - 88	4 1 - 2 - 3 - 5 - 16 - 18 - 20 - 20 - 11	13.0 12.6 12.3 12.9 12.0 11.2 11.1 11.0 11.1
304	FJ2	342	36-4 36-5 36-5	1600 730 900	27.5 27.1 27.3	. 04 . 02 . 00	7.1 7.2 7.2	- 4 - 17 - 20	81 80 76	20.0 18.5 18.2
305	FJ2	342	36-7 36-7 36-7 36-7 36-7	700 900 1140 1330 1450	27.0 27.3 27.4 27.4 27.4 27.4		7.2 7.1 7.1 7.1 7.1 7.1	- 22 - 27 - 31 - 36 - 38	71 63 58 51 50	16.9 16.3 15.5 15.1 14.8
306	FJ2	342	37-2 37-2	650 730	27.5 27.3	. 05 . 10	7. 3 7. 3	- 42 - 43	87 87	21.3 21.3
308	FJ2	483	38-6 38-6 38-6 38-6 38-7 38-7 38-7 38-7 38-7	700 900 1130 1240 1500 645 845 1100 1200	29. 2 29. 2 29. 4 29. 5 29. 5 30. 0 30. 0 30. 0 30. 0	. 02 . 02 . 02 . 02 . 02 . 02 . 02 . 02	7.55 7.55 7.55 7.55 7.55 7.55 7.55	- 92 - 95 -103 -104 -116 -115 -119 -118 -119	14 11 14 13 - 13 - 15 - 18 - 19	17.7 17.5 17.4 17.4 17.2 15.7 15.6 15.4 15.3
309	FJ2	483	39-2 39-2 39-2	545 645 750	30. 9 30. 9 30. 9	. 00 . 00 . 00	7.5 7.5 7.5	- 87 - 88 - 89	19 18 17	17.4 17.3 17.2
311	FJ2	605	40-3 40-3 40-3 40-3 40-4 40-4 40-4 40-4	715 920 1145 1305 1510 800 1015 1215 1430 1545 735 905	71.1 71.7 71.5 71.8 72.0 72.9 73.5 73.4 73.7 73.7 73.9 74.9 74.9	. 00 . 03 . 03 . 02 . 02 . 02 . 00 . 03 . 01 . 01 . 02 . 01	8.88 8.88 8.89 8.89 8.89 8.88 8.88 8.88	-184 -188 -101 -109 -105 -105 -103 -107 -111 -109 -109 -110	-119 -121 -123 -123 -123 -122 -117 -124 -125 -126 -129 -128	13.0 12.7 12.5 12.3 12.1 11.1 11.0 10.9 10.8 10.7 10.4 10.4
312	FJ2	605	40-7 40-7 40-7 40-7 40-7	710 900 1025 1140 1250	76.5 76.4 77.0 77.2 77.2	. 10 . 10 . 10 . 10 . 20	8.9 8.9 8.9 8.9 8.9 8.9	- 29 - 29 - 33 - 35 - 17	- 21 - 20 - 25 - 27 - 11	19.8 20.0 19.6 19.5 20.0
313	FJ2	605	41-2 41-2 41-2	755 1000 1100	78.9 79.0 78.9	. 10 . 10 . 10	8.9 8.9 8.9	- 13 - 20 - 21	- 12 - 20 - 23	19.9 19.6 19.5

Table 2

FJALLVEDEN - Field measurements

				The first of the set of		·				
Cali- bration nr	Bore- hole	Hole- length m	Date ww-d	Time	Conduc− tivity mS/m	Oxy- gen mg/l	рН	Eh,C	Eh,Pt mV	рS
202	FJ4	151	26-3 26-3 26-3 26-3 26-3 26-4 26-4 26-4 26-4 26-4 26-4 26-4 26-4	730 945 1140 1230 1335 1435 730 834 1035 1225 1300 1410 1510	$\begin{array}{cccccccccccccccccccccccccccccccccccc$		7.9 7.8 7.8 7.8 7.8 7.8 7.8 7.7 7.7 7.7	- 6 2 1 2 11 40 21 35 34 34 34 34 34 35	32 41 42 44 48 113 118 120 116 113 107 108	19.9 20.1 20.0 19.9 20.1 21.8 21.8 21.8 21.7 21.4 21.7 21.8 21.8
203	FJ4	151	26-6 26-6 26-7 26-7 26-7	1200 1420 1600 735 830	33.4 33.4 33.0 32.9 33.0	. 00 . 00 . 00 . 00 . 00		24 21 15 29 28	54 50 43 61 63	19.2 19.1 19.0 19.0 19.0
204	FJ4	151	27-1 27-1 27-1 27-1 27-1 27-1 27-2	1300 1430 1600 1700 1800 1200	35.0 35.1 35.2 35.3 35.4 35.3	. 04 . 05 . 04 . 04 . 04 . 04		- 83 - 87 - 85 - 83 - 88 - 82	- 62 - 64 - 61 - 59 - 62 - 57	13.2 12.6 12.4 12.4 12.1 12.1
204	FJ4	317	28-3 28-3 28-3 28-3 28-3 28-3 28-3 28-3	645 800 900 1200 1320 1430 1530 650	33. 4 33. 4 33. 4 33. 4 33. 5 33. 5 33. 5 33. 5 33. 4	00 00 01 01 02 02 02		- 76 - 79 - 81 - 84 - 85 - 86 - 88 - 125	- 17 - 18 - 19 - 23 - 25 - 26 - 26 - 115	13. 1 13. 0 12. 8 12. 7 12. 7 12. 7 12. 7
207	FJ4	317	28-6 28-6 28-6 28-6 28-7 28-7	1225 1400 1500 1635 750 920	33. 1 33. 2 33. 3 33. 3 33. 4 33. 7	. 04 . 04 . 04		- 24 - 25 - 21 - 20 - 28 - 26	- 16 - 14 - 11 - 12 - 19 - 17	19.4 18.8 18.9 18.9 18.0 17.8
208	FJ4	317	29-2 29-2 29-2 29-2	645 815 845 1000	34. 0 34. 0 34. 0 33. 9	. 00 . 01 . 02 . 03		- 75 - 83 - 85 - 90	3 1 1	14.6 14.5 14.5 14.4
209	FJ4	410	30-1 30-1	835 910	38. 1 38. 1	. 00 . 00		-178 -178	- 17 - 17	11.6 11.6
211	FJ4	410	30-6 30-6 30-6 30-6 30-6 30-7 30-7 30-7	635 800 1210 1400 800 900 930	31. 2 31. 2 31. 1 31. 2 31. 1 31. 5 31. 5 31. 5	. 00 . 03 . 05 . 05 . 05 . 00 . 00 . 00		-115 -118 -122 -130 -130 -156 -161 - 75	- 64 - 67 - 70 - 76 - 77 - 85 - 85 - 87	11.4 11.3 11.3 11.2 11.2 11.3 11.3 11.3
212	FJ4	410	31-1 31-2 31-2 31-2 31-2 31-2 31-2 31-2	1405 1530 745 745 840 930 1030 1130	32.0 32.0 31.8 31.8 31.9 31.9 31.9 31.9 32.0	. 00 . 00 . 00 . 00 . 00 . 01 . 03 . 04		-100 - 99 -109 -109 -109 -109 -109 -110 -109	- 89 - 89 - 84 - 84 - 84 - 84 - 85 - 85 - 86 - 87	13. 1 13. 0 12. 5 12. 5 12. 5 12. 5 12. 5 12. 4 12. 3

FJALLVEDEN - Field measurements

Table 3

Cali- bration nr	Bore- hole	Hole- length m	Date ww-d	Time	Conduc− tivity mS/m	Oxy- gen mg∕l	pH	Eh,C m∀	Eh,C ex mV	Eh,Pt mV	Eh,Pt ex mV	рS
214	FJ4	496	34-6 34-6 34-6 34-6 34-6 34-7 34-7	735 930 1145 1400 1500 1600 740 900	20.0 20.0 19.9 19.6 19.5 19.5 18.1 18.2			-171 -173 -173 -174 -174 -174 -194 -195	- 28 - 39 - 36 -219 -220 -221 -231 -233	-140 -144 -149 -156 -157 -158 -163 -165	- 43 - 39 - 47 -227 -228 -228 -235 -237	11.5 11.4 11.2 11.2 11.2 10.9 10.9
215	FJ4	496	35-1 35-1 35-1 35-2 35-2 35-2 35-2 35-2 35-2 35-2	1130 1340 1600 1740 800 955 1100 1245 1330	17.5 17.5 17.4 17.5 17.5 17.5 17.5 17.5 17.5	. 03 . 05 . 05 . 02 . 00 . 00 . 00 . 02 . 02 . 02		-160 -164 -165 -166 -167 -171 -171 -173 -173	-122 -128 -128 -129 -134 -140 -141 -141 -145	-154 -158 -159 -159 -159 -162 -163 -163 -164 -165	-142 -148 -155 -160 - 70 -102 -110 -117 -123	12.6 12.4 12.2 12.2 11.7 11.7 11.7 11.6 11.6
402	FJB	470	38-7 38-7 38-7 38-7 38-7	730 850 930 1205	23.8 23.5 23.8 23.8	. 01 . 01 . 03 . 02	8.4 8.4 8.4 8.4	-109 -109` -109 -110	- 65 - 67 - 66 - 75	-102 -103 -103 -103	- 82 - 85 - 85 - 94	16.4 16.3 16.3 16.3
403	FJB	470	39-1 39-2 39-2 39-2 39-2 39-2 39-2 39-2	1510 1430 745 930 1030 1215 1440	23. 9 23. 9 23. 9 23. 9 23. 9 23. 9 23. 9 23. 9	01 00 00 01 01 01	8.6 8.5 8.5 8.5 8.5 8.5 8.5	-118 -119 -119 -121 -121 -123 -124	- 86 - 87 - 96 - 99 -101 -105 -108	-113 -113 -113 -115 -116 -117 -119	-114 -116 -133 -137 -139 -143 -143	13.9 13.9 13.6 13.5 13.5 13.4 13.3
405	FJB	666	40-2 40-3 40-3 40-3 40-3 40-3 40-3 40-4 40-4	1230 1530 750 930 1100 1315 1530 750 1000 1200 1400 1550 750 900 1405	24. 0 24. 0	. 01 . 01 . 01 . 02 . 01 . 02 . 01 . 00 . 00 . 00 . 00 . 00 . 00 . 00	8.99 9.09 8.99 9.00 9.00 9.00 9.00 9.1 1 9.1	-153 -154 -155 -157 -158 -159 -160 -169 -169 -171 -173 -175 -177 -198 -199 -154	-156 -159 -176 -178 -181 -184 -201 -204 -205 -206 -207 -211 -212 -158	-140 -142 -141 -144 -145 -144 -145 -144 -143 -143 -143 -143 -143 -143 -143	-174 -180 -203 -205 -208 -211 -214 -228 -228 -229 -230 -235 -235 -177	12.7 12.4 12.4 12.3 12.3 12.3 12.3 12.2 12.1 12.1 12.1
406	FJB	666	40-6 40-6 40-6 40-7 40-7	1140 1400 1525 740 845	24. 0 24. 0 24. 0 24. 0 24. 0 24. 0	. 00 . 00 . 00 . 00 . 00	9.0 9.0 9.0 9.0 9.0 9.0	-136 -138 -138 -142 -142	-145 -150 -151 -165 -165	-128 -129 -129 -131 -131	-161 -167 -170 -192 -190	13.0 12.9 12.9 12.5 12.4
407	FJB	666	41-1 41-1 41-2 41-2 41-2 41-2 41-3	1300 1500 1600 750 935 1120 1430 800 915	24. 0 24. 0 24. 0 24. 0 24. 0 24. 0 24. 0 24. 1 24. 1	. 03 . 02 . 02 . 02 . 02 . 02 . 03 . 02 . 01 . 04	8.9 8.9 8.9 8.9 8.9 8.9 8.9 8.9 8.9 8.9	-141 -141 -143 -143 -144 -143 -144 -143 -141	-168 -169 -162 -184 -186 -188 -190 -199 -201	-132 -132 -132 -132 -132 -133 -134 -129 -130	-188 -189 -184 -207 -209 -211 -213 -223 -224	12.4 12.3 12.1 12.1 12.1 12.1 12.0 12.0 12.0

FJÄLLVEDEN - E⁰-values

Table 4

Bore- hole	Hole- length m	Cali- bration nr	Date ww-d	Glass pH4 mV	Glass pH7 mV	Glass pHiO mV	C pH4 mV	C pH7 mV	₽t pH4 mV	Pt pH7 mV	Ag,S O.O1M mV	Ag, S 0.05M mV
FJ2	123	301 302	34-4 34-7	391 385	396 388	400 395	283 259	259 257	278 256	256 257	-841 -845	-835 -835
FJ2	342	304 305 306	36-2 36-5 36-7	384 380 390	384 378 392	382 383 399	258 262 252	263 265 253	253 259 251	256 261 250	-822 -819 -829	-824 -814 -822
FJ2	483	307 308 309 310	37-2 38-4 38-7 39-2	382 379 378 484	384 383 376 482	391 387 407 390	263 258 265 264	255 257 259 259	253 255 256 258	248 255 259 256	-820 -809 -823 -817	-810 -805 -813 -809
FJ2	605	311 312 313 314	40-1 40-5 40-7 41-2	385 381 381 377	384 382 381 377	374 383 380 376	251 254 263 278	257 251 260 267	249 254 254 257	254 254 255 259	-820 -824 -827 -844	-823 -827 -832 -836
FJ4	151	201 202 203 204	25-2 26-1 26-5 26-7	392 389 383 453	395 383 381 470	389 383 397 483	250 251 252 259	249 253 254 257	246 244 251 255	247 248 253 256	-859 -836 -821 -809	-852 -827 -824 -816
FJ4	317	205 206 207 208	27-3 28-1 28-4 28-7	375 438 542 605	389 463 563 630	364 481 535 645	254 256 257 246	253 248 258 246	254 255 256 242	252 252 256 243	-827 -838 -835 -807	-824 -826 -833 -816
FJ4	410	209 210 211 212	29-3 30-1 30-4 30-7	563 806 342 233	592 761 528 409	610 810 575 558	250 244 251 251	251 244 237 240	250 243 249 250	251 244 243 240	-835 -839 -845 -868	-822 -830 -825 -854
FJ4	496	213 214 215 216	31-4 34-4 34-7 35-2	629 434 554 562	665 437 578 592	682 448 604 619	245 225 256 260	245 252 252 249	244 229 258 267	243 252 252 250	-878 -825 -841 -150	-857 -807 -824 - 47
FJB	470	401 402 403	38-2 38-5 38-7	394 384 388	413 386 390	400 388 392	253 264 251	282 246 250	253 269 247	282 244 247	-895 -838 -840	-873 -846 -846
FJ8	666	404 405 406 407 408	39-3 40-1 40-5 40-7 41-3	390 393 60 391 382	396 402 404 402 409	401 405 731 408 394	253 255 259 258 258	252 250 256 254 255	250 252 258 254 253	250 247 256 254 254	-836 -853 -836 -838 -831	-840 -841 -829 -829 -831

FJALLVEDEN - Field	i measurements	and	corresponding	laboratory	values
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Table 5a

Table 5b

Bore- hole	Depth	Hole- length m	Νr	Date sampling yy-ww-d	Date lab ww-d	pH field	pH 1≈b	Eh,C field mV	EH,₽t field mV	pS field	52- lab mq∕l	Cond. field mS∕m	Cond. lab mS∕m	Oxygen field mg∕l
				33										2.
FJ2	106	123												
			239	82-34-4	34-5	7.7	6.7	70	55	22.3	. 05	26	23.	. 30
			240	82-34-6	35-2	8.0	7.7	- 66	- 3	12.2	. 08	27	27	<. 02
			251	82-34-7	35-2	8.1	7.7	· 88	- 10	11.1	. 06	27	27	< 02
			252	82-35-1	35-2	8.0	7.7	5	15	19.2	. 06	27	27	. 05
FJ2	293	342												
102	E73	342	253	82-34-2	36-3		7.1				. 03		25	<. 02
			254	82-36-3	36-1	7.1	6.9	29	44	22.8	. 03	27	25	. 30
			255	82-36-4	36-5	7.1	6.9	2	81	20.6	. 04	28	25	. 00
			256	82-37-1	37-2	7.3	7.0	- 20	92	22.5	. 08	27	26	. 20
FJ2	409	483	~ ~ 7			7 0		05	00	23. 1	0.1	~ 7	05	70
			257	82-38-4	38-5	7.2	7.2	85	80 -209	17.4	.01	27 30	25 28	. 70
			258	82-38-6	39-2	7.5 7.5	7.3	-140	- 18	15.4	. 03	30	28	. 02 -
			259	82-38-7 82-39-1	39-2	7.3	7.4	- 40	43	20.1	. 02	30	20	. 02
			260	82-39-1	39-2		7.4	- 88	18	17.3	. 02	30 31	28	<. 02
			261	85-34-5	39-4	7.5	7.4	- 68	10	17.3	. 03	- 1	20	N. 02
FJ2	506	605												
			262	82-40-1	40-2	8.6	ፀ. ሪ	- 55	9	-22.3	. 02	66	63	. 40
			263	82-40-2	40-3	8.7	3.6	-125	- 25	20. 9	. 04	69	65	. 20
			264	82-40-3	40-4	8.8	9.6	-105	-123	12,4	. 11	72	68	. 03
			265	82-40-4	40-5	8.8	B. 7	1 07	-124	10.9	. 11	73	71	. 03
			266	82-40-7	41-2	8.9	8.8	- 34	- 26	19.5	. 01	77	73	. 10
			267	82-41-1	41-2	8.9	8.8	25	35	21.3	. 01	78	74	. 30
			268	82-41-2	41-4	8.9	9.7	- 20	- 20	19.6	< 01	79	72	. 10
			905				8.1						30	
			907				8.2						29	

FJALLVEDEN -	Field	measurements	and	corresponding	laboratoru	values

Bore- hole	Depth	Hole- length m	Nr	Date sampling yy-ww-d	Date lab ww-d	pH field	pH lab	Eh,C field mV	EH,Pt field mV	p5 field	S2- lab mg∕l	Cond. field mS∕m	Cond. lab mS∕m	Oxygen field mg∕l
4لF	131	151	211 212 213 214 215 216	82-26-2 82-26-3 82-26-4 82-26-7 82-27-1 82-27-1	26-3 26-4 26-5 27-2 27-2 27-3	7.9 7.8 7.7	7.7 7.7 7.5 7.5 8.2 7.8	2 1 36 - 83 - 34	22 42 107 - 62 - 8	20.8 19.9 21.8 13.2 14.7	. 04 . 04 . 02 . 05 . 05 . 05	34 35 33 35 35 34	33 34 32 35 34	. 05 . 04 . 03
FJ4	272	317	217 218 219 220 221 222 223	82281 82282 82283 82284 82287 82291 82292	28-2 29-3 28-4 28-5 29-2 29-2 29-3		7.6 7.5 7.5 7.5 7.5 7.5 7.4	- 47 - 86 - 90	- 23 - 26	13.6 12.7 15.2 14.4	03 02 04 02 < 01 01 03	33 33 34 33 34 34 34 34	30 32 32 33 33 33 33	. 10 . 03 . 02 . 04 . 02
¥ن¥	349	410	224 225 226 227 228 229 230	82-30-1 82-30-2 82-30-3 82-30-4 82-30-7 82-31-1 82-31-2	30-2 30-3 30-4 30-5 31-2 31-2 31-3		7, 6 7, 6 8, 1 9, 2 9, 3 9, 3 9, 3 9, 0	41 - 29 - 73 - 85 - 99 -110	49 - 18 - 54 -101 - 95 - 86	21.5 18.9 12.8 11.3 17.9 13.3 12.4	. 05 . 13 . 14 . 20 . 05 . 04 . 04	37 37 36 35 32 32 32	37 36 35 32 32 32	.50 .20 .10 .06 .03 <.02 .03
F ئ4	420	496	231 232 233 234 235 904 904	82-34-4 82-34-6 82-34-7 82-35-1 82-35-2	34-5 35-2 35-2 35-2 35-3		6.8 7.9 8.3 8.4 3.5 (3.1 (3.1	-174 -125 -160 -172	-156 -120 -154 -164	14.0 11.2 14.4 12.6 11.6	.05 .13 .13 .13 .13 .01	32 20 18 18 18	62 37 34 33 33 29 30	. 08 . 02 . 07 . 03 . 02

FJALLVEDEN - Field measurements and corresponding laboratory values

Table 5c

Bore- hole	Depth m	Hole- length m	Nr	Date sampling yy-ww-d	Date lab ww-d	pH field	pH lab	Eh,C field mV	EH,Pt field mV	pS field	52- lab mg/l	Cond. field mS∕m	Cond. lab mS∕m	Oxygen field mg/l
FJ8	402	470	281 282 283 284 285 286	82-38-3 82-38-4 82-38-5 82-38-7 82-39-1 82-39-2	38-4 38-5 39-2 39-3 39-3 39-4	8.0 8.3 8.4 8.4 8.5	7.6 7.7 7.9 7.9 8.0 7.9	- 53 - 5 - 97 -109 -117 -121	- 45 - 10 - 71 -103 -111 -116	18.2 21.0 16.8 16.3 14.0 13.5	. 05 . 03 . 01 . 01 . 02	23 23 24 24 24 24	22 22 22 22 22 22 22 22	04 20 04 03 02 < 02
FJ8	562	666	287 288 289 290 291 292 293	82-40-1 82-40-2 82-40-3 82-40-4 82-40-7 82-41-1 82-41-2	40-2 40-3 40-4 40-5 41-2 41-2 41-2	8.6 8.9 9.0 8.7 8.7 8.9 8.9	7.9 7.9 7.9 7.9 7.9 7.9 7.9 7.8	-134 -153 -159 -174 -106 -139 -144	-127 -140 -145 -143 -108 -130 -133	13.7 12.7 12.3 12.1 14.0 12.5 12.1	. 02 . 09 . 08 . 08 . 01 . 01 <. 01	24 24 24 24 24 24 24 24	22 22 22 22 22 22 22 22 22 22	02 < 02 < 02 < 02 04 03 03
			908 909				7.3 7.5						24 23	

Table 6a

FJÄLLVEDEN - Metal ions

Bore- hole	Depth m	Nr	Ca mg∕l	Mg mg∕l	Na mg∕l	K mg∕l	Mn mg∕l	Fe⊋+ mg/l	Fe-tot mg/l	Al mg∕l	Cu mg∕l	Sr mg∕l
FJ2	106											
		239	23	3.9	27	2.4	. 33	1.6	2.0	. 03	<. 005	. 124
		240	22	3.9	35	2.3	. 22	1.0	1.1	. 02	<. 005	. 128
		251	18	3.2	37	2.4	. 18	0.t	1.1	<. 01	<. 005	. 180
		252	18	3.0	37	2.4	. 18	. 95	1.1	<. 01	<. 005	. 125
FJ2	293											
		253	20	3.2	26	2.3	. 54	5.1	12			
		254	19	3.3	27	2.3	. 49	6.2	10			
		255	18	3.3	26	2.3	. 56	5.7	12			. 125
		256	19	3.3	26	2.3	. 49					. 122
FJ2	409											
		257	21	3.8	28	2.4	. 48	8.0	9.9	< 01	<. 005	. 125
		258	21	3.6	31	2.5	. 45	7.0	7.1			
		259	22	3.6	32	2.6	. 40	6.3	6.4			
		260	21	3.4	33	2.6	. 46	6.5	7.0	< 01	. 005	. 125
		261	21	3.6	34	2.6	. 40	4.7	5.5			
FJ2	506											
		262	13	1.2	113	1.2	. 08	. 42	. 83			
		263	13	1.2	117	1.2	. 08	. 32	. 83			
		264	13	1.3	122	1.2	. 07	. 42	. 52			
		265	13	1.0	125	1.0	. 07	. 40	. 48	<. 01	<. 005	. 125
		266	12	. 9	130	1.0	. 06	. 24	. 33			
		267	12	. 8	130	1.0	. 06	. 22	. 34	< 01	< 005	. 133
		268	12	1.2	129	1.0	. 06	. 42	. 42			
		905	31	5.0	26	4.7	. 06		. 10			
		907	13	2.9	49	2.6	. 10		. 27			

FUXLLVEDEN - Metal ions

Table 6b

Bore- hole	Depth m	NT	Ca mg∕l	Mg mg∕l	Naa mg∕ì	K mg∕l	Mrı mg∕l	Fe2+ mg∕l	Fe-tot mg/l	Al mg∕l	Cu mg∕l	Sr mg/l
FJ4	131											
		211	25	з. з	47	2.3	. 17	3.6	4.7			
		212	30	3.6	41	2.8	. 26	5.6	6.4			
		213	33	3.8	35	2.7	. 32	5.4	7.3	. 10	<. 005	192
		214	33	4.5	30	2.6	. 30	1.8	5.8	10	(00E	1 4 5
		215	15	2.2	65	3.0	. 10	1.1	1.3	. 10	< 005	. 145
		216	15	2.5	62	3.0	. 10	1.6	2.0			
FJ4	272											
		217	28	3.6	38	2.8	. 24	6. I	7.0			
		218	28	3.7	38	2.7	. 24	6.1	7.3			
		219	28	3.8	38	2.7	. 25	6. 5	7.5			
		220	28	3.8	38	2.6	. 25	6.4	7.3	. 05	<. 005	. 212
		221	28	3.7	39	2.7	. 27	6.4	7. B			
		222	28	3.9	38	2.7	. 30	B. 1	0 t	. 14	< 005	. 168
		223	29	3.9	37	2.7	. 29	8.2	9.8			
FJ4	349											
F 04	347	224	15	1.5	72	2.0	. 10	1.6	2.7			
		225	13	1.3	72	1.8	. 07	1.3	2.3			
		226	11	. 9	71	1.7	. 04	. 35	. 70			
		227	10	. 8	71	1.6	. 04	. 21	. 45	. 14	<. 005	. 072
		228	17	2.4	54	2.3	07	. 99	1.0			
		229	17	2.3	54	2.4	. 09	1.37	1.4	< 01	< 005	. 142
		230	17	2.3	56	2.5	. 07	1.78	1.8			
FJ4	420							F 1 F 1		07	< 00F	. 175
		231	21	2.0	100	2.8	. 28	3.3	11.0	. 07	<.005 <.005	. 107
		232	13	1.8	68	2.1	. 11	2.3	2.3	. 01 <. 01	<. 005	. 105
		233	13	1.8	62	2.0	01.	1.6	1.6		<. 005	. 113
		234	14	2.0	62	2.0	. 09	1.2	1.4	<. 01	< 005	. 115
		235	14	2.2	62	2.0	. 09	1.3	1.3	<. Ō1	< 005	. 110
		904	13	2.7	50	2.7	. 10		. 90			
		906	32	4.8	28	4.7	. 06		. 07			

							Table 6c					
				F	JALLVE.DE	EN - Meta	al ions					
Bore- hole	Depth m	Nr	Ca mg∕l	Mg mg∕l	Na mg∕1	K mg∕l	Mn mg∕l	Fe2+ mg∕l	Fe-tot mg∕l	Al mg∕l	Cu mg∕l	S⊤ mg⁄l
FJB	402											
		281	25	4.2	13	3.1	. 16	3.5	3. 8			
		282	25	4.4	13	3.1	. 14	2.1	2.5	<. 01	<. 005	. 091
		283	25	4.6	13	3.2	. 15	2.4	3.0			
		284	25	4.6	13	3.2	. 15					
		285	25	4.6	13	3.2	. 15	2.2	2.9			
		286	25	4.6	13	3.1	. 15	2.4	2. 9			
FJB	562											
		287	26	4.2	13	3.0	. 16	2.8	3.7			
		288	26	4.2	13	3.0	. 16	2.8	3.1			
		289	26	4.2	13	3.0	. 15	2.8	3.1			
		290	26	4.2	13	3.0	. 15	2.7	3. i	< 01	<. 005	. 091
		291	26	4.1	13	3.0	. 15	2.6	2.9			
		292	26	4.0	14	2.7	. 15	2.7	3.2	< 01	< 005	. 093
		293	26	4. 2	13	3.0	. 15	2.5	3.1			
							0.5		54			
		908	32	4.3	10	3.7	. 35		. 06			
		909	29	4.0	10	3.7	. 33		. 07			

FUALLVEDEN - Nitrogen containing ions

Bore-	Depth	Nr	N02	N03 mg/1	NH4 mg/l		N02-N M021	ND3-N mg/1	NH4−N ma∕l	Sum-N
hole	m		mg / 1	mg/i	mg / 1		my/i	mg/1	mg / I	mg∕1
FJ2	106									
		239	. 004	. 045	. 115		. 001	. 010	. 089	. 101
		240	. 005	. 020	. 085		. 002	. 005	. 066	072
		251	. 007	. 015	. 080		. 002	. 003	. 062	. 068
		252	<. 002	. 030	. 090		<. 001	. 007	. 070	. 077
FJ2	293	11	the second se	1.2.2	1.1			- 7		
		253	. 004	. 040	. 070		. 001	. 009	. 054	. 065
		254	<. 002	. 040	. 065		<. 001	. 009	. 050	. 060
		255	< 002	. 025	. 075		<. 001	. 006	. 058	. 064
		256	. 005	. 020	. 055		. 002	. 005	. 043	. 049
FJ2	400									
FUZ	409	057	007	. 030	050		. 002	007	000	0.10
		257	. 007		. 050		<. 001	. 007	. 039	. 048
		258	< 002	. 060	. 050		<. 001	. 014	. 039	. 053
		259	< 002	. 060	. 055	× ·	<. 001	. 014	. 043	. 057
		260	< 002	. 065	. 040		002	. 015	. 031	. 046
		261	. 005	. 035	. 065		. 00e	. 008	. 050	. 060
FJ2	506									
		262	. 004	. 040	. 015		. 001	. 009	.012	. 022
		263	. 005	. 030	. 010		. 002	. 007	. 008	. 016
		264	<. 002	. 030	. 010		<. 001	. 007	. 008	. 015
		265	<. 002	. 055	. 010		<. 001	. 012	. 008	. 021
		266	< 002	. 020	. 010		<. 001	. 005	. 008	. 013
		267	<. 002	. 030	. 010		<. 001	. 007	. 008	. 015
		268	<. 002	. 045	. 050		<. 001	. 010	. 039	. 050

Table 7b

FJALLVEDEN - Nitrogen containing ions

Bore- hole	Depth m	Nr	NO2 mg/1	ND3 mg/1	NH4 mg∕1	NO2-N mg/l	ND3−N mg / 1	NH4-N mg∕l	Sum−N mg⁄1
FJ4	131								
		211	. 005	. 040	. 120	. 002	. 009	. 093	. 104
		212	. 005	. 055	. 110	. 002	. 012	. 085	. 099
		213	. 095	4. 500	. 125	. 029	1.017	. 097	1.143
		214	. 520	1.570	. 230	. 158	. 355	. 179	. 692
		215	. 007	. 060	. 120	. 002	. 014	. 093	. 109
		216	. 027	. 025	. 260	. 008	. 006	. 202	. 216
FJ4	272								
		217	. 015	. 770	. 075	. 005	. 174	. 058	. 237
		218	. 015	. 065	. 135	. 005	. 015	. 105	. 124
		219	. 009	. 065	. 150	. 003	. 015	. 116	. 134
		220	. 030	. 015	. 045	. 009	003	. 035	. 047
		221	. 025	. 040	. 020	. 008	. 009	. 016	. 032
		222	. 030	. 030	. 120	·· . 009	. 007	. 093	. 109
		223	. 020	. 060	. 115	. 006	. 014	. 089	. 109
FJ4	349								
		224	. 012	. 025	. 010	. 004	. 006	. 008	. 017
		225	. 050	. 080	. 010	. 015	. 018	. 008	. 041
		226	. 004	. 010	. 020	. 001	. 002	. 016	. 019
		227	. 004	. 025	. 025	. 001	. 006	. 019	. 026
		228	. 004	. 035	. 075	. 001	. 008	. 058	. 067
		229	. 009	. 030	. 085	. 003	. 007	. 066	. 076
		230	. 004	. 045	. 065	. 001	. 010	. 050	. 062
FJ4	420								
		231	. 011	. 020	. 005	. 003	. 005	. 004	. 012
		232	. 012	. 020	. 020	. 004	. 005	. 016	. 024
		233	. 009	. 025	. 035	. 003	. 006	. 027	. 036
		234	. 007	. 040	. 050	. 002	. 009	. 039	. 050
		235	. 007	. 015	. 065	. 002	. 003	. 050	. 056

FJALLVEDEN - Nitrogen containing ions

Table 7c

Bore- hole	Depth m	Nr	NO2 mg/1	ND3 mg/1	NH4 mg/l	NO2-N mg/1	ND3-N mg/1	NH4-N mg/1	Sum-N mg∕l
FJ8	402								
		281	<. 002	. 040	. 015	<. 001	. 009	.012	. 021
		282	<. 002	015	. 010	<. 001	. 003	. 008	. 012
		283	<. 002	. 045	. 010	<. 001	. 010	. 008	. 019
		284	. 002	. 040	. 005	<. 001	. 009	. 004	. 014
		285	< 002	. 055	. 010	< 001	. 012	. 008	. 021
		286	. 005	. 020	. 015	. 002	. 005	. 012	. 018
FJ8	562								
		287	. 004	. 045	. 015	. 001	. 010	. 012	. 023
		288	. 004	. 020	. 010	. 001	. 005	. 008	. 014
		289	. 004	. 025	. 010	. 001	. 006	. 008	. 015
		290	. 012	. 015	. 015	. 004	. 003	. 012	. 019
		291	. 004	. 020	. 010	. 001	. 005	. 008	. 014
		292	. 005	. 015	. 010	. 002	. 003	. 008	. 013
		293	. 010	. 030	. 020	. 003	. 007	. 016	. 025

FJALLVEDEN - Remaining anions and other parameters

÷											
Bore- hole	Depth m	Nr	HCD3 mg∕l	C 1 mg ∕ 1	F mg/l	SO4 mg∕1	P04 mg/l	SiO2 mg∕1	\ TOC: mg≠1	Turb. NTU	Drilling water residue, I%
FJ2	106										
	100	239	130	8	. 3	10	. 045	14	6		7.9%
		240	160	6	. 7	7.5	. 050	13	5		2.2%
		251	160	5	. 7	6.1		13	5		1.7%
		252	160	5	. 7.	6.1	. 045	13	5		2.0%
EJ5	293										
		253	150	8	. 6	4.0	. 14	13	6		7.9%
		254	140	8	. 6	3.2	. 16	13	6	2.5	8.7%
		255	140	8	. 6	5.2	. 18	13	6	1.3	10.2%
		256	144	8	. 6	10	. 13	14	8	2.4	8.7%
FUR	100										
FUE	409	257	157	-7	. 7	~		13	5	13	7.1%
		258	171	7	. 9	. 2 . 2	. 22 . 18	13	5	9.7	8.7%
		259	173	8	. 9		. 16	13	6	6.0	8.7%
		260	170	8	. 9	. 2	. 20	13	5	28	7.9%
		261	175	9	. 9	. 2	. 14	13	4		8.7%
F J2	506										
		295	97	140	4.7	. 2	. 090	9.5	5	2.3	7.1%
		263	95	150	4.8	< 1	. 070	9.4	Ą	1.9	5.5%
		264	92	160	5.0	< 1	. 080	9.2	4	. 5	5 5%
		265	90	170	5.2	. 1	. 070	9.0	4	. 8 . 7	5.5%
		266	85	170	5.5	. 2	. 055	9.5	4		3.9%
		267	83	170	5.5	. 2	. 050	9.2	4	. 7	3.9%
		598	81	170	5.4	. 2	055	9.4	4	. 5	5, 5%
		905	180	4		7.9		15			
		907	178	4	. 7	7.9		15			
		707	1/0	4	. 7	7.0					

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FUXLLVEDEN - Remaining anions and other parameters

Table 8b

Table 8c

Bore-	Depth	Nr	нсоз	C 1	F	504	P04	Si02	тос	Turb.	Drilling water
hole	m		mg∕l	mg / 1	mg / 1	mg∕l	mg / 1	mg∕l	mg/1	NTU	residue: I%
FJ4	131										
		211	200	8	. 6	8	. 045	7.5	7	з. з	5.5%
		212	194	10	. 6	10	. 070	8.0	7	6.9	4.9%
		213	185	10	. 6	14	. 050	8.5	7	6.3	4.3%
		214	178	10	. 6	10	. 010	10	7	6.3	4.4%
		215	218	6	1.0	7	. 025	5.8	6	3.3	6.0%
		216	207	7	. 9	8	. 010	6.7	5	2.4	5.2%
FJ4	272										
101	turn / turn	217	192	9	. 6	8	. 085	8.3	a	6.8	4.1%
		218	190	9		7	. 075	8.3	8	9.4	4, 4%
		219	193	9	. 6	7	. 075	8.4	7	6.0	4. 4%
		220	191	9	. 6	5	. 130	8.5	н н	11	4.1%
		221	196	9	.5	5	. 055	8.1	7	6.9	4.7%
		222	196	9	. 5	7	. 045	8.2	7	2.2	4.4%
		223	196	9	. 6	7	. 035	8.4	7	5.1	4.4%
FJ4	349										
FQ 1	317	224	205	13	1.9	<. i	. 29	7.8	40	4.5	7.6%
		225	200	13	2.2	< 1	. 31	7.7	30	4.6	7.9%
		226	195	13	2.5	< 1	. 26	7.5	20	5.2	7.7%
		227	194	11	2.7	<. 1	. 24	7.5	15	5.7	7.9%
		228	194	5	1.1	2.6	. 115	6.7	6	1.5	3.4%
		229	195	5	1.0	3. 6	. 105	6.6		2.4	3.3%
		230	197	1	1.0	Э. 4	. 070	6.7	7	2.2	4.3%
FJ4	420										
		231	110	220	3.8	. 2	. 030	5.5	50		4.8%
		232	162	30	2.5	3.7	. 050	5.4	25		2.2%
		233	185	14	2.0	3.8	. 070	6. i	15		1.3%
		234	198	8	1.7	3.9	. 060	6.2	8		. 7%
		235	203	6	1.6	3.5	060	6.2	7		. 6%
			200	5		2. 0					
		904	184	4	. 9	7.3		14			
		904	184	4	. 7	7.9		15			
		700	104	-	. /	1.7		10			

FJALLVEDEN - Remaining anions and other parameters

Bore-	Depth	Nr	нсоз	C 1	F	S04	P04	Si02	TOC	Turb.	Drilling water
hole	m		mg∕l	mg / 1	mg∕1	mg∕l	mg / 1	mg∕l	mg / 1	NTU	residue, I%
FJB	402										
		281	123	4	1.0	7.5	. 015	7.8	3	4.4	5. 5%
		282	123	4	1.0	7	. 015	7.9	4	26	5. 5%
		283	128	4	1.0	6.5	. 020	8.0	3	20	6.3%
		284	128	4	1.0	6	. 020	8.3		22	
		285	130	4	1.0	6.5	. 025	8.4	3	22	5. 5%
		286	130	4	1.0	6	. 020	8.4	3	6.7	5. 5%
FJB	562										
		287	128	4	. 9	6	. 025	8.2	3	11	5. 5%
		288	130	4	. 9	6	. 030	8.2	З	6.0	5. 5%
		289	130	4	. 9	6	. 040	8.2	3	11.5	5. 5%
		290	130	4	. 9	6	. 045	8.7	з	9.0	4.7%
		291	132	4	. 9	6	. 050	8.4	3	11	3.9%
		292	130	4	. 9	5	. 050	8.5	3	11	4.7%
		293	129	4	. 9	5	. 055	8.5	з	8.5	7.1%
		908	140	3	. 4	6.9		18			
		909	137	3	. 4	7.0		17			

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Bore-	Depth	Nr	Sample	S, prt	Fe,prt	Al,prt
hole	m		ml	mg/l	mg∕l	mg∕l
FJ2	106	239	1900	. 002	. 074	. 095
FJ2	106	252	1700		. 032	. 112
FJ2	293	255	500	. 006	. 128	. 300
FJ2	293	256	500		. 118	. 220
FJ2	409	257	350	. 026	. 371	. 237
FJ2	409	260	1450	. 002	. 019	. 039
FJ2	506	265	2000	. 003	. 019	. 032
FJ2	506	267	950	. 004	. 103	. 021
FJ4 FJ4	131 131	213 215	600 850	012	. 133 . 032	. 267
FJ4	272	555	350	. 023	1.829	. 157
FJ4	272	550	425	. 009		. 329
FJ4 FJ4	349 349	227 229	300 2250	047	270	. 667 . 017
FJ4 FJ4	420 420	231 234	1200 1950	018	. 242	. 258 . 042
FUB	402	282	2250	. 012	. 356	008
FJ8	562	290	2250	. 004	. 016	004
FJ8	562	292	2250	. 004	. 013	

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Table 10

FJALLVEDEN - Uranium, thorium, radium and radon

Bore-	Depth	Nr	Th	U	Ra-226	Rn-222
hole	m		ug/l	Bq/l	Bq/1	Bq/1
FJ2	106	252	(2.6+0.6)E-2	(1.4+0.4)E-2	(5.5+0.8)E-3	201+5
FJ2	293	256	(9.2+4.5)E-2	(6.1+3.2)E-3	(5.5+1.0)E-3	266+5
FJ2	409	260	(1.2+0.4)E-1	< 2.3 E-3	< 5.8 E-3	120+5
FJ2	506	267	< 2.6 E-1	< 5.1 E-3	(3.5+0.3)E-3	305+6
FJ4	131	215	<pre>< 1.2 E-2 (4.4+1.8)E-3 (1.9+0.6)E-2 (2.5+0.6)E-2</pre>	(9.3+0.5)E-2	< 3.9 E-3	151+5
FJ4	272	222		(7.7+0.5)E-2	< 3.0 E-3	240+5
FJ4	349	229		(1.6+0.3)E-2	(8.5+1.3)E-3	166+5
FJ4	420	234		< 4.8 E-3	< 1.1 E-3	189+5
FJ8	402	285	< 3.8 E-2	(2.8+0.4)E-2	(2. 9+0. 6)E-3	144+5
FJ8	562	292	< 6.9 E-2	(1.1+0.3)E-2	(3. 1+0. 8)E-3	107+5

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